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COMPONENTS: Control (1) artis: Cu.0; [1137-30-1] CONTINUE (1) artis: Cu.0; [1247-30-1] Charlon (1) artis: Sucrements: Sucrements: Sucrements: Sucrements: Control (1) (124-38-9] Control (1) artis: Concentration of chlorides, pressure of CO2, and temperature. Concentration of chlorides, pressure of CO2, and temperature. Sait Count of a control (1) artis: Solubility of Cu.20 in chloride solutions. Sait Count of a control (1) artis: Solubility of Cu.20 in chloride solutions. Sait Count of a control (1) artis: Solubility of Cu.20 in chloride solutions. Sait Count of a control (1) artis: Solubility of Cu.20 and temperature. Sait Count of a control (1) artis: Solubility of Cu.20 and (1) artis: Count of a control (1) artis: Count of a cont of a control (1) artis: Count of a control (copper(I) Oxide	
Concentration of chlorides, pressure of CO2, and temperature. T. P. Dirkse EXFERIMENTAL VALUES: Solubility of Cu20 in chloride solutions. Cu2/s dm 3 " Cu2/s dm 4 " Cu2/s dm 4 NUMENTION </th <th> Copper (2) Sodium Cl Potassium Carbon d </th> <th>hlordie; NaCÍ; [7647- m chloride; KCl; [744 ioxide; CO₂ [124-38-9</th> <th>14-5] 7~40-7]</th> <th>Shlyapnikov, D. 8 <i>Khim.</i> <u>1977</u>, 22,</th> <th>5.; Shtern, E. K. Zh. Neorg , 1100-6; Russ. J. Inorg.</th>	 Copper (2) Sodium Cl Potassium Carbon d 	hlordie; NaCÍ; [7647- m chloride; KCl; [744 ioxide; CO ₂ [124-38-9	14-5] 7~40-7]	Shlyapnikov, D. 8 <i>Khim.</i> <u>1977</u> , 22,	5.; Shtern, E. K. Zh. Neorg , 1100-6; Russ. J. Inorg.
CO2, and temperature. EXPERIMENTAL VALUES: Solubility of Cu2O in chloride solutions. Salt $C_{salt}/mol dm^{-3}$ $20^{\circ}C$, $pCO_2 = 50$ atm $200^{\circ}C$, $pCO_2 = 90$ atm Macl 1 2.4127 4.8735 " 2 5.4181 10.2503 " 3 8.2000 15.0500 " 3 8.2000 19.1500 KCl 1 4.7449 5.8305 " 3 13.0541 21.8021 " 3 13.0541 21.8021 " 3 13.0541 21.8021 was maintained at 4 mol dm ⁻³ by adding NANO. The solubility of Cu_0O increased with in-creasing Ci ion concentration but no numerical data are given. Tw so found that the cation also had an effect on the solubility of Cu_0O. Again, no numerical data are given. Muer these conditions, with the other alkali metal ions being less effective. The Cu_QO and solvent were placed in an autor-clave Which was shaken mechanically for 24 Inverted and the mixture was filtered through a corundum filter fitted into the autoclave was anitained by adding a calculated amount of solid CO2. ESTIMATED ERROR: No details are given. No details are given. ESTIMATED ERROR:					
$\frac{Salt}{none} \frac{C_{GL}/g}{mone} \frac{20^{\circ}C, pC_{Q_{2}} = 50 \text{ atm}}{C_{GL}/g} \frac{20^{\circ}C, pC_{Q_{2}} = 90 \text{ atm}}{C_{Q_{2}}/g} \frac{20^{\circ}C, pC_{Q_{2}}}{15.000} \frac{20^{\circ}C}{15.000} \frac{20^{\circ}C}{15.000} \frac{20^{\circ}C}{16.000} \frac{20^{\circ}C}{16.000} \frac{20^{\circ}C}{16.000} \frac{20^{\circ}C}{16.000} \frac{20^{\circ}C}{16.0000} \frac{20^{\circ}C}{16.0$			sure of	T. P. Dirkse	
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