

COMPONENTS:		ORIGINAL MEASUREMENTS:												
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Sapozhnikova, N. V.; Postovskii, I. Ya.												
(2) Water; H ₂ O; [7732-18-5]		Zh. Prikl. Khim 1944, 17, 427-34.												
VARIABLES:		PREPARED BY:												
Temperature		R. Piekos												
EXPERIMENTAL VALUES:														
<table><tr><td rowspan="2">t/°C</td><td colspan="2">Solubility</td></tr><tr><td>Weight%</td><td>10² mol kg⁻¹ water^a</td></tr><tr><td>20</td><td>0.627</td><td>2.94</td></tr><tr><td>37</td><td>1.16</td><td>5.48</td></tr></table>				t/°C	Solubility		Weight%	10 ² mol kg ⁻¹ water ^a	20	0.627	2.94	37	1.16	5.48
t/°C	Solubility													
	Weight%	10 ² mol kg ⁻¹ water ^a												
20	0.627	2.94												
37	1.16	5.48												
^a Calculated by compiler.														
AUXILIARY INFORMATION														
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:												
Sulfacetamide was dissolved in water to form a satd soln which was occasionnally agitated in a glass vessel immersed in a thermostat. The equilibrium was usually attained after 1 h. Five- to 100-cm ³ samples of the satd soln were placed in Pt crucibles or dishes and evapd to dryness at temps lower than 110-115°C. The residue was dried to const wt at 105-110°C and weighed.		Pure, recrystd sulfacetamide was used. Its mp conformed to that reported in the literature.												
		Purity of the water was not specified.												
		ESTIMATED ERROR:												
		Soly: quite reliable results were obtained over the temp range 20-75°C (authors).												
		Temp: ±0.05°C (authors).												
		REFERENCES:												

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9] (2) Water; H ₂ O; [7732-18-5]		ORIGINAL MEASUREMENTS: Langecker, H. <i>Arch. Exptl. Path. Pharmacol.</i> <u>1948</u> , <u>205</u> , 291-301.	
VARIABLES: One temperature: 37°C		PREPARED BY: R. Piekos	
EXPERIMENTAL VALUES: Solubility of sulfacetamide in water at 37°C is 1400 mg% (6.53 x 10 ⁻² mol dm ⁻³ , compiler).			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE: An excess of sulfacetamide in water was boiled for 1 h in a sealed ampul followed by keeping the soln at 37°C. Before assaying, the solute was treated with a 2.6N NaOH soln (1) to cleave the acetyl group and the sulfanilamide was detd colorimetrically by the method of Bratton and Marshall (2) using a Havemann colorimeter (3), as well as by microanal detn of the solid residue.		SOURCE AND PURITY OF MATERIALS: Source and purity of sulfacetamide was not specified. The water was free of oxidants.	
		ESTIMATED ERROR: Nothing specified.	
		REFERENCES: 1. Scudi, J. V. <i>J. Lab. Clin. Med.</i> <u>1940</u> , <u>25</u> , 404. 2. Bratton, A. G.; Marshall, E. K. <i>J. Biol. Chem.</i> <u>1939</u> , <u>128</u> , 537. 3. Havemann R. <i>Klin. Wochenschr.</i> <u>1940</u> , p. 503.	

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Gusyakov, V. P.; Likholt'ot, N. M. <i>Farm. Zh. (Kiev)</i> <u>1960</u> , <i>15</i> (3), 21-4.
VARIABLES: One temperature: 20°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in water at 20°C is 0.501 g/100 g water (2.34×10^{-2} mol kg^{-1} , compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: A small excess of sulfacetamide was equilibrated for 8 h in a 50-ml test tube with 20 ml of water. Aliquots were taken with a pipet fitted with a filter. Sulfacetamide was detd at 295 nm using a SF-4 spectrophotometer.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was prepd from a purified Na salt by neutralizing it with equivalent quantity of aq HCl. The product was repeatedly washed with water and its purity conformed to the requirements of the State Pharmacopeia VIII. Purity of the water was not specified.
	ESTIMATED ERROR: Soly: the accuracy corresponded to that of the colorimetric detns (authors). Temp: not specified.
	REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likhol'ot, N. M. <i>Farm. Zh. (Kiev)</i> <u>1965</u> , 20(5), 44-6.
VARIABLES: One temperature: 20°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in water at 20°C is 0.501 g/100 ml (2.34×10^{-2} mol dm ⁻³ , compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: An earlier described method was employed (1) whereby a small excess of sulfacetamide was equilibrated with 20 ml of water for 8 h in a 50-ml test tube. Aliquots were withdrawn through a filter and sulfacetamide was assayed bromatometrically.	SOURCE AND PURITY OF MATERIALS: Nothing specified. ESTIMATED ERROR: Soly: not specified. Temp: $\pm 0.1^\circ C$ (authors). REFERENCES: 1. Gusyakov, V. P.; Likhol'ot, N. M. <i>Farm. Zh. (Kiev)</i> <u>1960</u> , 15(8), 21.

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Yamazaki, M.; Aoki, M.; Kamada, A.; Yata, N. <i>Yakuzaiigaku</i> <u>1967</u> , <i>27(1)</i> , 37-40.
VARIABLES: One temperature: 30°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in water at 30°C is 41.4 mmol/L (8.87 g dm ⁻³ , compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: Sulfacetamide (0.5 g) was placed in an L-shaped tube together with 20 ml of water. The mixt was shaken in a thermostat until equilibrium was attained. The sulfaceta- mide was then assayed in the supernatant spectrophotometrically at 545 nm on a Beckmann DU spectrophotometer. The results were taken from a calibration graph.	SOURCE AND PURITY OF MATERIALS: Nothing specified.
	ESTIMATED ERROR: Soly: not specified. Temp: $\pm 1^\circ C$ (authors).
	REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Water; H_2O [7732-18-5]	ORIGINAL MEASUREMENTS: Gusyakov, V. P.; Likhol'ot, N. M.; Kutna, I. M. <i>Farm. Zh. (Kiev)</i> <u>1967</u> , <i>22</i> (3), 34-9.
VARIABLES: One temperature: 20°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in water at 20°C is 0.50 g/100 ml (2.3×10^{-2} mol dm ⁻³ , compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: An excess of sulfacetamide in water was equilibrated for 24 h in an ampul immersed in a water thermostat. Aliquots of the satd soln were withdrawn through a filter and the sulfacetamide content was assayed in the filtrate photometrically.	SOURCE AND PURITY OF MATERIALS: Purity of the sulfacetamide conformed to the requirements of the State Pharmacopeia IX. Purity of the water was not specified.
	ESTIMATED ERROR: Soly: not specified. Temp: $\pm 0.1^\circ C$ (authors).
	REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Gusyakov, V. P.; Likholt'ot, N. M.; Kutna, I. M. <i>Farm. Zh. (Kiev)</i> <u>1968</u> , 23(6), 56-61.
VARIABLES: One temperature: 21-25°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in water at room temperature (21-25°C) is 0.501 g/100 ml (2.334×10^{-2} mol dm ⁻³ , compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: Small quantities (2-4 mg) of sulfacetamide were added to a known quantity of water under stirring until satn was attained.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide: neither source nor purity was specified. Purity of the water was not specified.
	ESTIMATED ERROR: Nothing specified.
	REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Shkadova, A. I. <i>Farm. Zh. (Kiev)</i> 1969, 24(3), 39-41.
VARIABLES: One temperature: 20°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in water at 20°C is 3.02×10^{-2} mol/kg (0.647 g/100 g, compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: A satd aqueous soln of sulfacetamide was equilibrated in a water thermostat at $20 \pm 0.1^\circ C$. The concn of sulfacetamide was detd by alkalimetric titrn.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was prepd from its Na salt by addn of equivalent quantity of 0.1N HCl. The product was washed with water and dried. Distd. water was used.
	ESTIMATED ERROR: Soly: not specified. Temp: $\pm 0.1^\circ C$ (author).
	REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Rohdewald, P. <i>Pharm. Ztg.</i> 1971, No. 38 1342-4.
VARIABLES: One temperature: 20°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: <p>Solubility of sulfacetamide in water at 20°C is 0.318_g b/50 ml ($2.975_8 \times 10^{-2}$ mol dm⁻³, compiler) and 0.315 g/50 ml (2.93×10^{-2} mol dm⁻³, compiler)^a.</p> <p>^a Two values were given without comment (compiler).</p>	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: The soln was equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.	SOURCE AND PURITY OF MATERIALS: The source and purity of the materials was not specified.
	ESTIMATED ERROR: Soly: mean std deviation (68.3% of results deviating by S g), $S = 0.019$; relative std deviation 6.09%; no of measurements 34 (author). Temp: $\pm 0.05^\circ C$ (author).
	REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Kaneniwa, N.; Watari, N.; Iijima, H. <i>Chem. Pharm. Bull.</i> <u>1978</u> , <i>26</i> (9), 2603-14.
VARIABLES: One temperature: 37°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in water at 37°C is 12.0 mg/ml solution (5.60×10^{-2} mol dm ⁻³ , compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: An excess of sulfacetamide was placed in a flask contg 25 ml of water. The flask was shaken (2 strokes/s at the amplitude of 3 cm) in a thermostatically controlled water bath at 37°C. One-ml sample was withdrawn every 6 h (total equilibration period was 3-5 days) using a warmed Millipore filter syringe with a filter pore size of 0.45μ (Millipore HAWP 01300) and the filtrate was dild with water and assayed spectrophotometrically (1).	SOURCE AND PURITY OF MATERIALS: Comm sulfacetamide of the Japanese Pharmacopeia grade and distd water were used. ESTIMATED ERROR: Soly: not specified. Temp: $\pm 0.05^\circ C$ (authors). REFERENCES: 1. Kanenisa, N.; Watari, N. <i>Chem. Pharm. Bull.</i> <u>1974</u> , <i>22</i> , 1969.

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)sulfonyl]-(sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Lithium chloride; LiCl; [7447-41-8] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M., Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.														
VARIABLES: Concentration of LiCl	PREPARED BY: R. Piekos														
EXPERIMENTAL VALUES: <div data-bbox="291 561 1016 1100"> <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of LiCl (mol kg⁻¹)</th> <th>Solubility at 20°C (mol kg⁻¹)</th> </tr> </thead> <tbody> <tr> <td>0</td> <td>2.5</td> </tr> <tr> <td>0.25</td> <td>3.0</td> </tr> <tr> <td>0.5</td> <td>2.9</td> </tr> <tr> <td>1.0</td> <td>2.7</td> </tr> <tr> <td>1.5</td> <td>2.6</td> </tr> <tr> <td>2.0</td> <td>2.5</td> </tr> </tbody> </table> </div>		Concentration of LiCl (mol kg ⁻¹)	Solubility at 20°C (mol kg ⁻¹)	0	2.5	0.25	3.0	0.5	2.9	1.0	2.7	1.5	2.6	2.0	2.5
Concentration of LiCl (mol kg ⁻¹)	Solubility at 20°C (mol kg ⁻¹)														
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1.0	2.7														
1.5	2.6														
2.0	2.5														
AUXILIARY INFORMATION															
METHOD/APPARATUS/PROCEDURE: Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a LiCl soln was placed and a small excess of sulfacetamide. The mixts were equilibrated at 20°C for 18 h. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was purified by crystn. LiCl was purified by a recommended procedure (1). The source and purity of the materials were not specified. ESTIMATED ERROR: Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors). REFERENCES: 1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye khimicheskiye reaktivy</i> , Moscow, 1955.														

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [133-80-9] (2) Sodium chloride; NaCl; [7647-14-5] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Langecker, H. <i>Arch. Exptl. Path. Pharmacol.</i> <u>1948</u> , 205, 291-301.
VARIABLES: One temperature: 37°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: <p>Solubility of sulfacetamide in physiological saline (0.9% w/w NaCl solution) at 37°C is 983 mg% (4.59×10^{-2} mol dm⁻³, compiler).</p>	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: <p>An excess of sulfacetamide was added to physiological saline and boiled for 1 h in a sealed ampul followed by keeping the ampul at 37°C. Before assaying, the solute was treated with a 2.6 N NaOH soln (1) to cleave the acetyl group and the sulfanilamide was detd colorimetrically by the method of Bratton and Marshall (2) using a Havemann colorimeter (3), as well as by microanal detn of the solid residue.</p>	SOURCE AND PURITY OF MATERIALS: Source and purity of the materials was not specified. ESTIMATED ERROR: Nothing specified. REFERENCES: 1. Scudi, J. V. <i>J. Lab. Clin. Med.</i> <u>1940</u> , 25, 404. 2. Bratton, A. G.; Marshall, E. K. <i>J. Biol. Chem.</i> <u>1939</u> , 128, 537. 3. Havemann, R. <i>Klin. Wochenschr.</i> <u>1940</u> , p. 503.

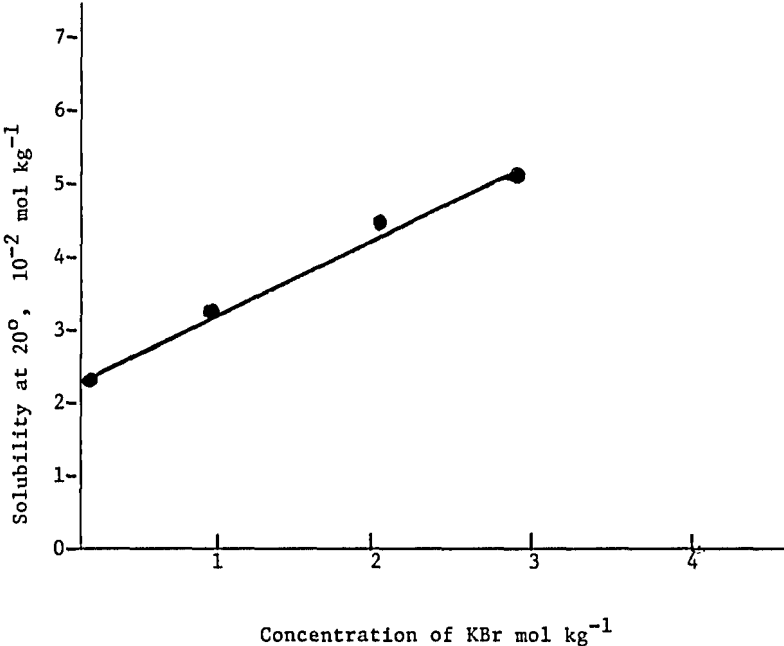
COMPONENTS: (1) Acetamide, N - [(4-aminophenyl)sulfonyl- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Sodium chloride; NaCl; [7647-14-5] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M.; Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.														
VARIABLES: Concentration of NaCl	PREPARED BY: R. Piekos														
EXPERIMENTAL VALUES: <div data-bbox="305 524 1029 990" data-label="Figure"> <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of NaCl (mol kg⁻¹)</th> <th>Solubility at 20°C (10² mol kg⁻¹)</th> </tr> </thead> <tbody> <tr><td>0.1</td><td>2.5</td></tr> <tr><td>0.2</td><td>2.8</td></tr> <tr><td>0.5</td><td>2.7</td></tr> <tr><td>1.0</td><td>2.5</td></tr> <tr><td>1.5</td><td>2.3</td></tr> <tr><td>2.0</td><td>2.0</td></tr> </tbody> </table> </div>		Concentration of NaCl (mol kg ⁻¹)	Solubility at 20°C (10 ² mol kg ⁻¹)	0.1	2.5	0.2	2.8	0.5	2.7	1.0	2.5	1.5	2.3	2.0	2.0
Concentration of NaCl (mol kg ⁻¹)	Solubility at 20°C (10 ² mol kg ⁻¹)														
0.1	2.5														
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1.5	2.3														
2.0	2.0														
AUXILIARY INFORMATION															
METHOD/APPARATUS/PROCEDURE: Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a NaCl soln was placed and a small excess of sulfacetamide. The mixts were equilibrated for 18 h at 20°C. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was purified by crystn. NaCl was purified by a recommended procedure (1). The source and purity of the materials were not specified. ESTIMATED ERROR: Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors). REFERENCES: 1. Karyakin, Ya. V. ; Angelov, I. I. <i>Chistye khimicheskyye reaktivy</i> , Moscow, 1955.														

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Phosphoric acid, disodium salt; Na_2HPO_4 ; [7558-94-4] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Krüger-Thiemer, E. <i>Arch. Dermatol. Syphilis</i> 1942, 183, 90-116.
VARIABLES: One temperature: ea 20°C; one pH: 8.74	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: <p>Solubility of sulfacetamide in a 0.705M (10%) Na_2HPO_4 solution of pH 8.74 at room temperature (about 20°C) is 5.230 g% (2.441×10^{-1} mol dm^{-3} solution, compiler).</p>	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: Sulfacetamide (0.5 g) was dissolved in the 0.705 M (10%) Na_2HPO_4 soln (pH 8.74) at room temp (about 20°C), shaken for 2 h at 20°C, and filtered. A 1 cm ³ aliquot was then withdrawn, cooled, 1 cm ³ of 2N HCl was added, and the sulfacetamide content was detd colorimetrically by the Marshall method modified by Kimmig (1) using an Authenreith colorimeter. The pH was detd on an ultraionograph using a glass electrode.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was the product manufd by Schering AG under the name Albucid. The source and purity of the remaining materials was not specified. ESTIMATED ERROR: Soly: precision $\pm 5\%$ (author). Temp: not specified. pH: ± 0.05 pH unit (author). REFERENCES: 1. Kimmig, J. <i>Arch. Dermatol.</i> 1938, 176, 722; <i>Erg. Hyg.</i> 1941, 24, 398.

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Gusyakov, V. P.; Likhol'ot, N. M.	
(2) Potassium chloride; KCl; [7447-40-7]		Farm. Zh. (Kiev) 1960, 15(3), 21-4.	
(3) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of KCl		R. Piekos	
EXPERIMENTAL VALUES:			
Concentration of KCl		Solubility at 20°C	
Weight %	g/100 g water	10 ² mol kg ⁻¹ ^a	
0.74	0.493	2.30	
1.82	0.551	2.57	
3.59	0.566	2.64	
6.93	0.541	2.53	
12.97	0.454	2.12	
15.70	0.394	1.84	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
A small excess of sulfacetamide was equilibrated for 8 h in a 50-ml test tube with 20 ml of aqueous KCl soln. Aliquots were taken with a pipet fitted with a filter. Sulfacetamide was detd in the filtrate at 285 nm using a SF-4 spectro-photometer.		Sulfacetamide was prepd from a purified Na salt by neutralizing it with equivalent quantity of aq. HCl. The product was repeatedly washed with water and conformed to the requirements of the State Pharmacopeia VIII. KCl was doubly crystd. Purity of the water was not specified.	
		ESTIMATED ERROR:	
		Soly: the accuracy corresponded to that of colorimetric detns (authors).	
		Temp: not specified.	
		REFERENCES:	

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl-)sulfonyl]-(sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Potassium chloride; KCl; [7447-40-7] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M.; Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.														
VARIABLES: Concentration of KCl	PREPARED BY: R. Piekos														
EXPERIMENTAL VALUES: <div data-bbox="308 572 1050 1103"> <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of KCl (mol kg⁻¹)</th> <th>Solubility at 20°C (10² mol kg⁻¹)</th> </tr> </thead> <tbody> <tr><td>0.0</td><td>2.5</td></tr> <tr><td>0.25</td><td>3.0</td></tr> <tr><td>0.5</td><td>3.0</td></tr> <tr><td>1.0</td><td>3.0</td></tr> <tr><td>1.5</td><td>3.0</td></tr> <tr><td>2.0</td><td>3.0</td></tr> </tbody> </table> </div>		Concentration of KCl (mol kg ⁻¹)	Solubility at 20°C (10 ² mol kg ⁻¹)	0.0	2.5	0.25	3.0	0.5	3.0	1.0	3.0	1.5	3.0	2.0	3.0
Concentration of KCl (mol kg ⁻¹)	Solubility at 20°C (10 ² mol kg ⁻¹)														
0.0	2.5														
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2.0	3.0														
AUXILIARY INFORMATION															
METHOD/APPARATUS/PROCEDURE: Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a KCl soln was placed and a small excess of sulfacetamide. The mixts were equilibrated at 20°C for 18 h. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotomer.	<div data-bbox="742 1287 1274 1461"> SOURCE AND PURITY OF MATERIALS: Sulfacetamide was purified by crystn. KCl was purified by a recommended procedure (1). The source and purity of the materials were not specified. </div> <div data-bbox="742 1614 1274 1747"> ESTIMATED ERROR: Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors). </div> <div data-bbox="742 1747 1274 1952"> REFERENCES: 1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye khimicheskiye reaktivy.</i> Moscow, 1955. </div>														

COMPONENTS:		ORIGINAL MEASUREMENTS:																						
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Gusyakov, V. P.; Likholt'ot, N. M.																						
(2) Potassium bromide; KBr; [7758-02-3]		Farm. Zh. (Kiev) 1960, 15(3), 21-4.																						
(3) Water; H ₂ O; [7732-18-5]																								
VARIABLES:		PREPARED BY:																						
Concentration of KBr		R. Piekos																						
EXPERIMENTAL VALUES:																								
<table><tr><th>Concentration of KBr</th><th colspan="2">Solubility at 20°C</th></tr><tr><th>Weight %</th><th>g/100 g water</th><th>10² mol kg⁻¹^a</th></tr><tr><td>1.17</td><td>0.501</td><td>2.34</td></tr><tr><td>2.88</td><td>0.514</td><td>2.40</td></tr><tr><td>5.61</td><td>0.587</td><td>2.74</td></tr><tr><td>10.63</td><td>0.668</td><td>3.12</td></tr><tr><td>19.22</td><td>0.769</td><td>3.59</td></tr></table>				Concentration of KBr	Solubility at 20°C		Weight %	g/100 g water	10 ² mol kg ⁻¹ ^a	1.17	0.501	2.34	2.88	0.514	2.40	5.61	0.587	2.74	10.63	0.668	3.12	19.22	0.769	3.59
Concentration of KBr	Solubility at 20°C																							
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^a Calculated by compiler.																								
AUXILIARY INFORMATION																								
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:																						
A small excess of sulfacetamide was equilibrated for 8 h in a 50-ml test tube with 20 ml of aqueous KBr soln. Aliquots were taken with a pipet fitted with a filter. Sulfacetamide was detd in the filtrate at 285 nm using a SF-4 spectrophotometer.		Sulfacetamide was prepd from a purified Na salt by neutralizing it with equivalent quantity of aq HCl. The product was repeatedly washed with water and conformed to the requirements of the State Pharmacopeia VIII. KBr was doubly crystd. Purity of the water was not specified.																						
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		Temp: not specified.																						
		REFERENCES:																						

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)sulfonyl] (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Potassium bromide; KBr; [7758-02-3] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M.; Gussyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.										
VARIABLES: Concentration of KBr	PREPARED BY: R. Piekos										
EXPERIMENTAL VALUES:  <table border="1" data-bbox="315 506 1094 1154"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of KBr (mol kg⁻¹)</th> <th>Solubility at 20°C (10⁻² mol kg⁻¹)</th> </tr> </thead> <tbody> <tr> <td>0.2</td> <td>2.3</td> </tr> <tr> <td>0.8</td> <td>3.2</td> </tr> <tr> <td>1.8</td> <td>4.4</td> </tr> <tr> <td>2.8</td> <td>5.1</td> </tr> </tbody> </table>		Concentration of KBr (mol kg ⁻¹)	Solubility at 20°C (10 ⁻² mol kg ⁻¹)	0.2	2.3	0.8	3.2	1.8	4.4	2.8	5.1
Concentration of KBr (mol kg ⁻¹)	Solubility at 20°C (10 ⁻² mol kg ⁻¹)										
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AUXILIARY INFORMATION											
METHOD/APPARATUS/PROCEDURE: Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a KBr soln was placed and a small excess of sulfacetamide. The mixts were equilibrated at 20°C for 18 h. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was purified by crystn. KBr was purified by a recommended procedure (1). The source and purity of the reagents were not specified. ESTIMATED ERROR: Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors). REFERENCES: 1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye khimicheskiye reaktivy</i> , Moscow, 1955.										

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Potassium iodide; KI; [7681-11-0] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M; Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.								
VARIABLES: Concentration of KI	PREPARED BY: R. Piekos								
EXPERIMENTAL VALUES: <div data-bbox="329 538 921 893" data-label="Figure"> <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of KI (mol/kg)</th> <th>Solubility at 20°C (10² mol/kg)</th> </tr> </thead> <tbody> <tr> <td>0.2</td> <td>2.0</td> </tr> <tr> <td>1.0</td> <td>3.3</td> </tr> <tr> <td>2.0</td> <td>5.0</td> </tr> </tbody> </table> </div>		Concentration of KI (mol/kg)	Solubility at 20°C (10² mol/kg)	0.2	2.0	1.0	3.3	2.0	5.0
Concentration of KI (mol/kg)	Solubility at 20°C (10² mol/kg)								
0.2	2.0								
1.0	3.3								
2.0	5.0								
AUXILIARY INFORMATION									
METHOD/APPARATUS/PROCEDURE: <p>Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a KI soln was placed and a small excess of sulfacetamide. The mixts were equilibrated at 20°C for 18 h. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.</p>	SOURCE AND PURITY OF MATERIALS: <p>Sulfacetamide was purified by crystn. KI was purified by a recommended procedure (1). The source and purity of the materials were not specified.</p> ESTIMATED ERROR: <p>Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors).</p> REFERENCES: <p>1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye khimicheskiye reaktivy</i>, Moscow, 1955.</p>								

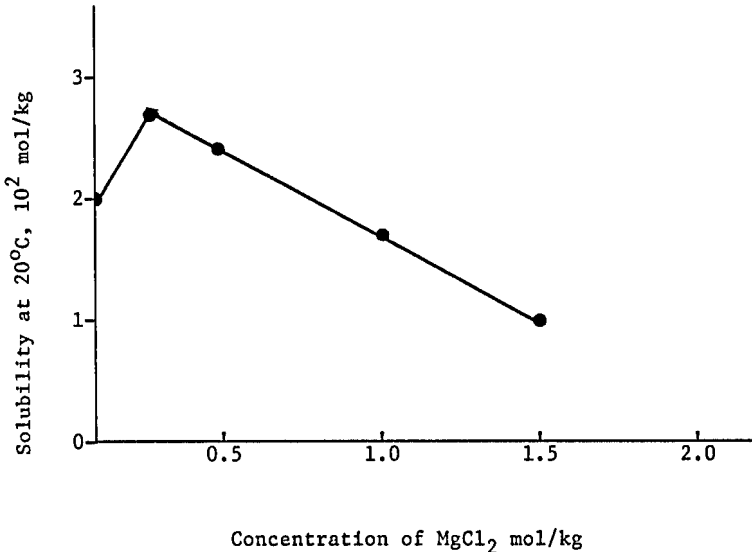
COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Potassium iodide; KI; [7681-11-0] (3) Water; H_2O [7732-18-5]	ORIGINAL MEASUREMENTS: Gussyakov, V. P.; Likhol'ot, N. M. <i>Farm. Zh. (Kiev)</i> <u>1960</u> , 15(3), 21-4.																		
VARIABLES: Concentration of KI	PREPARED BY: R. Piekos																		
EXPERIMENTAL VALUES: <table border="1" data-bbox="366 499 1038 749"> <thead> <tr> <th data-bbox="366 499 618 528">Concentration of KI</th> <th colspan="2" data-bbox="730 499 954 528">Solubility at 20°C</th> </tr> <tr> <th data-bbox="422 532 534 560">Weight %</th> <th data-bbox="632 532 800 560">g/100 g water</th> <th data-bbox="856 532 1010 560">$10^2 \text{ mol kg}^{-1}{}^a$</th> </tr> </thead> <tbody> <tr> <td data-bbox="450 600 506 628">1.63</td> <td data-bbox="674 600 744 628">0.597</td> <td data-bbox="898 600 954 628">2.79</td> </tr> <tr> <td data-bbox="450 633 506 661">3.98</td> <td data-bbox="674 633 744 661">0.615</td> <td data-bbox="898 633 954 661">2.87</td> </tr> <tr> <td data-bbox="450 665 506 693">7.66</td> <td data-bbox="674 665 744 693">0.752</td> <td data-bbox="898 665 954 693">3.51</td> </tr> <tr> <td data-bbox="436 697 520 725">14.23</td> <td data-bbox="674 697 744 725">0.843</td> <td data-bbox="898 697 954 725">3.94</td> </tr> </tbody> </table> <p data-bbox="417 762 716 802">^a Calculated by compiler.</p>		Concentration of KI	Solubility at 20°C		Weight %	g/100 g water	$10^2 \text{ mol kg}^{-1}{}^a$	1.63	0.597	2.79	3.98	0.615	2.87	7.66	0.752	3.51	14.23	0.843	3.94
Concentration of KI	Solubility at 20°C																		
Weight %	g/100 g water	$10^2 \text{ mol kg}^{-1}{}^a$																	
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AUXILIARY INFORMATION																			
METHOD/APPARATUS/PROCEDURE: A small excess of sulfacetamide was equilibrated for 8 h in a 50-ml test tube with 20 ml of aqueous KI soln. Aliquots were taken with a pipet fitted with a filter. Sulfacetamide was detd in the filtrate at 285 nm using a SF-4 spectrophotometer.	<table border="1"> <tr> <td data-bbox="730 1249 1285 1576"> SOURCE AND PURITY OF MATERIALS: Sulfacetamide was prepd from a purified Na salt by neutralizing it with equivalent quantity of aq HCl. The product was repeatedly washed with water and conformed to the requirements of the State Pharmacopeia VIII. KI was doubly crystd. Purity of the water was not specified. </td> </tr> <tr> <td data-bbox="730 1580 1285 1709"> ESTIMATED ERROR: Soly: the accuracy corresponded to that of colorimetric detns (authors). Temp: not specified. </td> </tr> <tr> <td data-bbox="730 1713 1285 1923"> REFERENCES: </td> </tr> </table>	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was prepd from a purified Na salt by neutralizing it with equivalent quantity of aq HCl. The product was repeatedly washed with water and conformed to the requirements of the State Pharmacopeia VIII. KI was doubly crystd. Purity of the water was not specified.	ESTIMATED ERROR: Soly: the accuracy corresponded to that of colorimetric detns (authors). Temp: not specified.	REFERENCES:															
SOURCE AND PURITY OF MATERIALS: Sulfacetamide was prepd from a purified Na salt by neutralizing it with equivalent quantity of aq HCl. The product was repeatedly washed with water and conformed to the requirements of the State Pharmacopeia VIII. KI was doubly crystd. Purity of the water was not specified.																			
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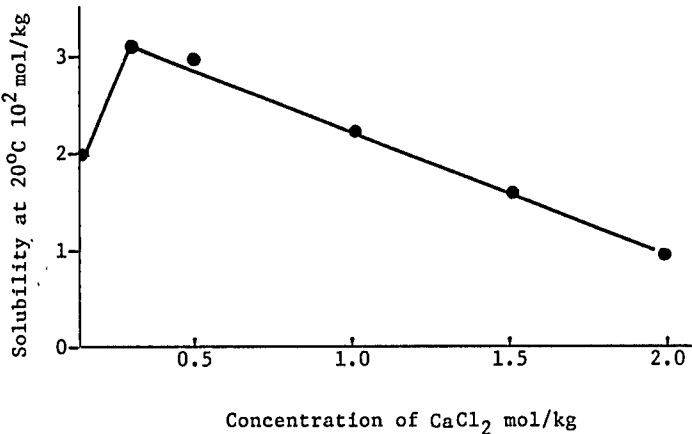
COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Gusyakov, V. P.; Likholt'ot, N. M.	
(2) Potassium thiocyanate; KSCN; [333-20-0]		Farm. Zh. (Kiev) 1960, 15(2), 21-4.	
(3) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of KSCN		R. Piekos	
EXPERIMENTAL VALUES:			
Concentration of KSCN		Solubility at 20°C	
Weight %	g/100 g water	10 ² mol kg ⁻¹ ^a	
0.96	0.582	2.72	
2.37	0.700	3.27	
4.63	0.796	3.72	
8.85	1.096	5.12	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
A small excess of sulfacetamide was equilibrated for 8 h in a 50-ml test tube with 20 ml of aqueous KSCN soln. Aliquots were taken with a pipet fitted with a filter. Sulfacetamide was detd in the filtrate at 285 nm using a SF-4 spectro-photometer.		Sulfacetamide was prepd from a purified Na salt by neutralizing it with equivalent quantity of aq HCl. The product was repeatedly washed with water and conformed to the requirements of the State Pharmacopeia VIII. KSCN was doubly crystd. Purity of the water was not specified.	
		ESTIMATED ERROR:	
		Soly: the accuracy corresponded to that of colorimetric detns (authors) Temp: not specified.	
		REFERENCES:	

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)sulfonyl]-(sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Potassium thiocyanate; KCNS; [333-20-0] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M.; Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.										
VARIABLES: Concentration of KCNS	PREPARED BY: R. Piekos										
EXPERIMENTAL VALUES: <div data-bbox="340 516 900 1030"> <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of KCNS (mol/kg)</th> <th>Solubility at 20°C (10² mol/kg)</th> </tr> </thead> <tbody> <tr> <td>0.2</td> <td>2.0</td> </tr> <tr> <td>1.0</td> <td>4.3</td> </tr> <tr> <td>1.5</td> <td>5.4</td> </tr> <tr> <td>2.0</td> <td>7.2</td> </tr> </tbody> </table> </div>		Concentration of KCNS (mol/kg)	Solubility at 20°C (10 ² mol/kg)	0.2	2.0	1.0	4.3	1.5	5.4	2.0	7.2
Concentration of KCNS (mol/kg)	Solubility at 20°C (10 ² mol/kg)										
0.2	2.0										
1.0	4.3										
1.5	5.4										
2.0	7.2										
AUXILIARY INFORMATION											
METHOD/APPARATUS/PROCEDURE: <p>Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a KCNS soln was placed and a small excess of sulfacetamide. The mixts were equilibrated at 20°C for 18 h. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.</p>	SOURCE AND PURITY OF MATERIALS: <p>Sulfacetamide was purified by crystn. KCNS was purified by a recommended procedure (1). The source and purity of the materials were not specified.</p> ESTIMATED ERROR: <p>Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors).</p> REFERENCES: <p>1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye khimicheskiye reaktivy</i>, Moscow, <u>1955</u>.</p>										

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Phosphoric acid, monopotassium salt; KH_2PO_4 ; [7778-77-0] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Krüger-Thiemer, E. <i>Arch. Dermatol. Syphilis</i> <u>1942</u> , 183, 90-116.
VARIABLES: One temperature: ca 20°C; one pH: 4.37	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in a 0.735M (10%) KH_2PO_4 solution of pH 4.37, at room temperature (about 20°C), is 0.632 g% (2.95×10^{-2} mol dm ⁻³ solution, compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: Sulfacetamide (0.5 g) was dissolved in the 0.735M (10%) KH_2PO_4 solution (pH 4.37) at room temp (about 20°C), shaken for 2 h at 20°C, and filtered. A 2 cm ³ aliquot was then withdrawn, cooled, 1 cm ³ of 2N HCl was added, and the sulfacetamide content was detd colorimetrically by the Marshall method modified by Kimmig (1) using an Autenrieth colorimeter. The pH was detd on an ultraionograph using a glass electrode.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was the product manufd by Schering AG under the name Albucid. The source and purity of the remaining materials was not specified. ESTIMATED ERROR: Soly: precision ± 5% (author). Temp: not specified. pH: ±0.05 pH unit (author). REFERENCES: 1. Kimmig, J. <i>Arch. Dermatol.</i> <u>1938</u> , 176, 722; <i>Erg. Hyg.</i> <u>1941</u> , 24, 398.

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)sulfonyl]-(sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Ammonium chloride; NH_4Cl ; [12125-02-9] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M.; Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.														
VARIABLES: Concentration of NH_4Cl	PREPARED BY: R. Piekos														
EXPERIMENTAL VALUES: <div data-bbox="301 544 1083 1068"> <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of NH_4Cl (mol/kg)</th> <th>Solubility at 20°C (10^2 mol/kg)</th> </tr> </thead> <tbody> <tr><td>0.1</td><td>2.0</td></tr> <tr><td>0.2</td><td>3.0</td></tr> <tr><td>0.5</td><td>3.0</td></tr> <tr><td>1.0</td><td>3.0</td></tr> <tr><td>1.5</td><td>3.0</td></tr> <tr><td>2.0</td><td>3.0</td></tr> </tbody> </table> </div>		Concentration of NH_4Cl (mol/kg)	Solubility at 20°C (10^2 mol/kg)	0.1	2.0	0.2	3.0	0.5	3.0	1.0	3.0	1.5	3.0	2.0	3.0
Concentration of NH_4Cl (mol/kg)	Solubility at 20°C (10^2 mol/kg)														
0.1	2.0														
0.2	3.0														
0.5	3.0														
1.0	3.0														
1.5	3.0														
2.0	3.0														
AUXILIARY INFORMATION															
METHOD/APPARATUS/PROCEDURE: <p>Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a NH_4Cl soln was placed and a small excess of sulfacetamide. The mixts were equilibrated for 18 h at 20°C. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.</p>	SOURCE AND PURITY OF MATERIALS: <p>Sulfacetamide was purified by crystn. NH_4Cl was purified by a recommended procedure (1). The source and purity of the materials were not specified.</p> ESTIMATED ERROR: <p>Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors).</p> REFERENCES: <p>1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye khimicheskiye reaktivy</i>, Moscow, 1955.</p>														

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)sulfonyl]-(sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Magnesium chloride; $MgCl_2$; [7786-30-3] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M.; Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.												
VARIABLES: Concentration of $MgCl_2$	PREPARED BY: R. Piekos												
EXPERIMENTAL VALUES:  <table border="1" data-bbox="225 513 969 1064"> <caption>Experimental Data Points</caption> <thead> <tr> <th>Concentration of $MgCl_2$ (mol/kg)</th> <th>Solubility at 20°C (10^2 mol/kg)</th> </tr> </thead> <tbody> <tr> <td>0</td> <td>2.0</td> </tr> <tr> <td>0.2</td> <td>2.7</td> </tr> <tr> <td>0.5</td> <td>2.4</td> </tr> <tr> <td>1.0</td> <td>1.7</td> </tr> <tr> <td>1.5</td> <td>1.0</td> </tr> </tbody> </table>		Concentration of $MgCl_2$ (mol/kg)	Solubility at 20°C (10^2 mol/kg)	0	2.0	0.2	2.7	0.5	2.4	1.0	1.7	1.5	1.0
Concentration of $MgCl_2$ (mol/kg)	Solubility at 20°C (10^2 mol/kg)												
0	2.0												
0.2	2.7												
0.5	2.4												
1.0	1.7												
1.5	1.0												
AUXILIARY INFORMATION													
METHOD/APPARATUS/PROCEDURE: Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a $MgCl_2$ soln was placed and a small excess of sulfacetamide. The mixts were equilibrated for 18 h at 20°C. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide was purified by crystn. $MgCl_2$ was purified by a recommended procedure (1). The source and purity of the materials were not specified. ESTIMATED ERROR: Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors). REFERENCES: 1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye khimicheskiye reaktivy</i> , Moscow, 1955.												

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Calcium chloride; $CaCl_2$; [10043-52-4] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M.; Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.														
VARIABLES: Concentration of $CaCl_2$	PREPARED BY: R. Piekos														
EXPERIMENTAL VALUES:  <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of $CaCl_2$ (mol/kg)</th> <th>Solubility at 20°C (10^2 mol/kg)</th> </tr> </thead> <tbody> <tr><td>0.1</td><td>2.0</td></tr> <tr><td>0.2</td><td>3.1</td></tr> <tr><td>0.5</td><td>3.0</td></tr> <tr><td>1.0</td><td>2.3</td></tr> <tr><td>1.5</td><td>1.6</td></tr> <tr><td>2.0</td><td>1.0</td></tr> </tbody> </table>		Concentration of $CaCl_2$ (mol/kg)	Solubility at 20°C (10^2 mol/kg)	0.1	2.0	0.2	3.1	0.5	3.0	1.0	2.3	1.5	1.6	2.0	1.0
Concentration of $CaCl_2$ (mol/kg)	Solubility at 20°C (10^2 mol/kg)														
0.1	2.0														
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0.5	3.0														
1.0	2.3														
1.5	1.6														
2.0	1.0														
AUXILIARY INFORMATION															
METHOD/APPARATUS/PROCEDURE: <p>Satd soln of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a $CaCl_2$ soln was placed and a small excess of sulfacetamide. The mixts were equilibrated for 18 h at 20°C. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.</p>	SOURCE AND PURITY OF MATERIALS: <p>Sulfacetamide was purified by crystn. $CaCl_2$ was purified by a recommended procedure (1). The source and purity of the materials were not specified.</p> ESTIMATED ERROR: <p>Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors).</p> REFERENCES: <p>1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye Khimicheskiye reaktivy</i>, Moscow, 1955.</p>														

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Barium chloride; $BaCl_2$; [10361-37-2] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholet, N. M.; Gusyakov, V. P. <i>Med. Prom. SSSR</i> <u>1963</u> , 17(5), 28-31.												
VARIABLES: Concentration of $BaCl_2$	PREPARED BY: R. Piekos												
EXPERIMENTAL VALUES: <div data-bbox="257 506 1002 956"> <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>Concentration of $BaCl_2$ (mol/kg)</th> <th>Solubility at 20°C (10^2 mol/kg)</th> </tr> </thead> <tbody> <tr> <td>0.0</td> <td>2.1</td> </tr> <tr> <td>0.25</td> <td>3.1</td> </tr> <tr> <td>0.5</td> <td>2.8</td> </tr> <tr> <td>1.0</td> <td>2.2</td> </tr> <tr> <td>1.5</td> <td>1.5</td> </tr> </tbody> </table> </div>		Concentration of $BaCl_2$ (mol/kg)	Solubility at 20°C (10^2 mol/kg)	0.0	2.1	0.25	3.1	0.5	2.8	1.0	2.2	1.5	1.5
Concentration of $BaCl_2$ (mol/kg)	Solubility at 20°C (10^2 mol/kg)												
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AUXILIARY INFORMATION													
METHOD/APPARATUS/PROCEDURE: <p>Satd solns of sulfacetamide were prepd in 50-ml tightly closed ampuls in which 20 ml of a $CaCl_2$ soln was placed and a small excess of sulfacetamide. The mixts were equilibrated for 18 h at 20°C. Aliquots were pipetted out through a filter, dild, and assayed spectrophotometrically at 285 nm on a SF-IV spectrophotometer.</p>	SOURCE AND PURITY OF MATERIALS: <p>Sulfacetamide was purified by crystn. $BaCl_2$ was purified by a recommended procedure (1). The source and purity of the materials were not specified.</p> ESTIMATED ERROR: <p>Soly: measurements were repeated several times (authors). Temp: $\pm 0.1^\circ C$ (authors).</p> REFERENCES: <p>1. Karyakin, Ya. V.; Angelov, I. I. <i>Chistye khimicheskiye reaktivy</i>, Moscow, 1955.</p>												

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Likhol'ot, N. M. <i>Farm. Zh. (Kiev)</i> 1965, 20(5), 44-6.	
(2) Phosphoric acid, monosodium salt; NaH ₂ PO ₄ ; [7558-80-7]			
(3) Potassium chloride; KCl; [7447-40-7]			
(4) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of NaH ₂ PO ₄ - KCl		R. Piekos	
EXPERIMENTAL VALUES:			
Concentration of NaH ₂ PO ₄ -KCl ^a		Solubility of sulfacetamide at 20°C	
mol/l		g/100 ml	10 ² mol dm ⁻³ ^b
0.088		0.602	2.05
0.112		0.597	2.03
0.139		0.597	2.03
0.165		0.597	2.03
0.182		0.596	2.03
0.185		0.596	2.03
^a KCl was added in such amounts as to correct the ionic strength of solution.			
^b Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
An earlier described method was employed (1) whereby a small excess of sulfacetamide was equilibrated with 20 ml of a NaH ₂ PO ₄ -KCl soln for 8 h in a 50-ml test tube. Aliquots were withdrawn through a filter and sulfacetamide was assayed bromatometrically.		Nothing specified.	
		ESTIMATED ERROR:	
		Soly: not specified.	
		Temp: $\pm 0.1^{\circ}\text{C}$ (authors).	
		REFERENCES:	
		1. Gusyakov, V. P.; Likhol'ot, N. M. <i>Farm. Zh. (Kiev)</i> 1960, 15(8), 21.	

COMPONENTS:				ORIGINAL MEASUREMENTS:			
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9]				Krüger-Thiemer, E.			
(2) Phosphoric acid, disodium salt; Na_2HPO_4 ; [7558-94-4]				Arch. Dermatol. Syphilis 1942, 183, 90-116.			
(3) Phosphoric acid, monopotassium salt; KH_2PO_4 ; [7778-77-0]							
(4) Water; H O; [7732-18-5]							
VARIABLES:				PREPARED BY:			
Temperature, pH				R. Piekos			
EXPERIMENTAL VALUES:							
Composition of 1/15M phosphate buffer solutions				Solubility			
				Room temp (ca 20°C)		37°C	
Na_2HPO_4	KH_2PO_4	%Content	pH	g% solution ^a	10^2 mol dm^{-3}	g% solution ^a	10^2 mol dm^{-3}
1.0	99.0	0.91	4.944	0.830	3.874	-	-
10.0	90.0	0.91	5.906	0.909	4.242	1.220	5.694
61.1	38.9	0.93	7.005	1.632	7.617	1.770	8.261
9.5	0.5	0.733 ^b	7.51	4.710	21.98	-	-
94.7	5.3	0.95	0.018	2.232	10.42	-	-
^a Calculated by compiler.							
^b Molar content; 10% buffer solution.							
AUXILIARY INFORMATION							
METHOD/APPARATUS/PROCEDURE:				SOURCE AND PURITY OF MATERIALS:			
Sulfacetamide (0.5 g) was dissolved in 10 cm ³ of a buffer soln, shaken for 2 h at 20°C. (or left for 48 h at 37°C), and filtered at respective temp. A 1 cm ³ aliquot was then withdrawn, cooled, 1 cm ³ of 2N HCl was added, and the sulfacetamide content was detd colorimetrically by the Marshall method modified by Kimmig (1) using an Authenreith colorimeter. The pH values were detd on an ultraionograph using a glass electrode.				Sulfacetamide was the product manufd by Schering AG under the name Albucid. The source and purity of the remaining material was not specified.			
				ESTIMATED ERROR:			
				Soly: precision $\pm 5\%$ (author). Temp: not specified. pH: ± 0.05 pH unit (author).			
				REFERENCES:			
				1. Kimmig, J. Arch. Dermatol. 1938, 176, 722; Erg. Hyg. 1941, 24, 398.			

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Langecker, H.	
(2) Phosphoric acid, disodium salt; Na ₂ HPO ₄ ; [7558-94-4]		Arch. Exptl. Path. Pharmacol. <u>1948</u> ,	
(3) Phosphoric acid, monopotassium salt; KH ₂ PO ₄ ; [7778-77-0]		205, 291-301.	
(4) Wafer; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
pH		R. Piekos	
EXPERIMENTAL VALUES:			
pH of the 1/15M phosphate buffer		Solubility at 37°C	
	mg%	10 ² mol dm ⁻³ ^a	
4.9	978	4.56	
5.9	974	4.56	
6.9	1607	7.50	
7.5	2241	10.46	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
An excess of sulfacetamide was added to the buffer soln and boiled for 1 h in a sealed ampul followed by keeping the ampul at 37°C. Before assaying the solute was treated with a 2.6N NaOH soln (1) to cleave the acetyl group and the sulfanilamide was detd colorimetrically by the method of Bratton and Marshall (2) using a Havemann colorimeter (3), as well as by microanal detn of the solid residue.		Source and purity of the materials was not specified.	
		ESTIMATED ERROR:	
		Nothing specified.	
		REFERENCES:	
		1. Scudi, J. V. <i>J. Lab. Clin. Med.</i> <u>1940</u> , 25, 404.	
		2. Bratton, A. G.; Marshall, E. K. <i>J. Biol. Chem.</i> <u>1939</u> , 128, 537.	
		3. Havemann, R. <i>Klin. Wochenschr.</i> <u>1940</u> , p. 503.	

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Phosphoric acid, disodium salt; Na_2HPO_4 ; [7558-94-4] (3) Phosphoric acid, monopotassium salt; KH_2PO_4 ; [7778-77-0] (4) Water; H_2O ; [7732-18-5]		ORIGINAL MEASUREMENTS: Bandelin, F. J.; Malesh, W. <i>J. Am. Pharm. Assoc., Sci. Ed.</i> 1959, 48, 177-81.	
VARIABLES: pH		PREPARED BY: R. Piekos	
EXPERIMENTAL VALUES: Solubility of sulfacetamide in buffers of varying mixtures of $Na_2HPO_4 \cdot 7H_2O$ (71.6 g/l distilled water; 0.27 mol dm^{-3} , compiler) and KH_2PO_4 (36.3 g/l distilled water; 0.27 mol dm^{-3} , compiler) at 37°C			
Initial pH		Solubility	
		mg/100 ml	mol dm^{-3} ^a
	5.0	1250	0.0583
	5.5	1350	0.0629
	6.0	2150	0.100
	6.5	3020	0.141
	7.0	4400	0.205
	7.5	15,000	0.699
	8.0	41,000	1.911
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE: Solns were prepd by adding an excess of sulfacetamide to 10 ml of buffer soln at each pH level in 18 x 150-mm test tubes, stoppering the tubes, and placing them in a water bath at 37°C with gentle agitation for 24 h. The mixt was then filtered and a 1-ml aliquot was accurately pipetted into a volumetric flask for diln and analysis. The balance was retained for pH detn to ascertain any change in pH value. Sulfacetamide was assayed colorimetrically by the method of Bratton and Marshall as described in detail by Biamonte and Schneller (1). A standard curve was prepd using accurately prepd standard solutions.		SOURCE AND PURITY OF MATERIALS: Neither source nor purity of the reagents were specified. Distilled water was used.	
		ESTIMATED ERROR: Soly: av values of duplicate runs are reported (authors). Temp and pH: not specified.	
		REFERENCES: 1. Biamonte, A. R.; Schneller, G. E. <i>J. Am. Pharm. Assoc., Sci. Ed.</i> , 1952, 41, 341.	

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Phosphoric acid, disodium salt; Na_2HPO_4 ; [7558-94-4] (3) Phosphoric acid, monopotassium salt; KH_2PO_4 ; [7778-77-0] (4) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Yamazaki, M.; Aoki, M.; Kamada, A.; Yata, N. <i>Yakuzaigaku</i> <u>1967</u> , <i>27</i> (1), 37-40.
VARIABLES: One temperature: 30°C; one pH: 7.4	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in a phosphate buffer solution of pH 7.4 ^a ($\mu = 0.17$) at 30°C is 91.0 mmol/L (19.50 g dm ⁻³ , compiler). ^a At the end of experiment the pH was 5.6.	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: Sulfacetamide (0.5 g) was placed in an L-shaped tube together with 20 ml of a buffer soln. The mixt was shaken in a thermostat until equilibrium was attained. The sulfacetamide was then assayed in the supernatant spectrophotometrically at 545 nm on a Beckmann DU spectrophotometer. The results were taken from a calibration graph.	SOURCE AND PURITY OF MATERIALS: Nothing specified. ESTIMATED ERROR: Soly and pH: not specified. Temp: $\pm 1^\circ C$ (authors). REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Phosphoric acid, disodium salt; Na_2HPO_4 ; [7558-94-4] (3) 1,2,3-Propanetricarboxylic acid, 2-hydroxy- (citric acid); $C_6H_8O_7$; [77-92-91] (4) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Likholt, N. M. <i>Farm. Zh. (Kiev)</i> 1965, 20(5), 44-6.																							
VARIABLES: pH	PREPARED BY: R. Piekos																							
EXPERIMENTAL VALUES: <table border="1" data-bbox="203 537 941 858"> <thead> <tr> <th data-bbox="203 537 541 623" rowspan="2">pH of McIlvaine's buffer solution</th><th colspan="2" data-bbox="546 537 941 572">Solubility at 20°C</th></tr> <tr> <th data-bbox="546 578 679 623">g/100 ml</th><th data-bbox="686 578 941 623">$10^2 \text{ mol dm}^{-3}{}^a$</th></tr> </thead> <tbody> <tr> <td>4.1</td><td>0.627</td><td>2.93</td></tr> <tr> <td>5.1</td><td>0.884</td><td>4.13</td></tr> <tr> <td>5.9</td><td>1.605</td><td>7.49</td></tr> <tr> <td>6.5</td><td>2.502</td><td>11.68</td></tr> <tr> <td>6.9</td><td>3.140</td><td>14.66</td></tr> <tr> <td>7.5</td><td>3.585</td><td>16.73</td></tr> </tbody> </table> <p data-bbox="303 889 607 921">^a Calculated by compiler.</p>		pH of McIlvaine's buffer solution	Solubility at 20°C		g/100 ml	$10^2 \text{ mol dm}^{-3}{}^a$	4.1	0.627	2.93	5.1	0.884	4.13	5.9	1.605	7.49	6.5	2.502	11.68	6.9	3.140	14.66	7.5	3.585	16.73
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AUXILIARY INFORMATION																								
METHOD/APPARATUS/PROCEDURE: An earlier described method was employed (1) whereby a small excess of sulfacetamide was equilibrated with 20 ml of the McIlvaine's buffer soln for 8 hr in a 50-ml test tube. Aliquots were removed through a filter and sulfacetamide was assayed bromatometrically.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide: not specified. McIlvaine's buffer solns were prepd from a 0.2M Na_2HPO_4 and a 0.1M citric acid solns. Source and purity of the buffer components were not specified. ESTIMATED ERROR: Soly: not specified. Temp: $\pm 0.1^\circ C$ (authors). pH: not specified. REFERENCES: 1. Gussyakov, V. P.; Likholt, N. M. <i>Farm. Zh. (Kiev)</i> 1960, 15(8), 21.																							

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide) $C_8H_{10}N_2O_3S$: [144-80-9] (2) Calcium chloride; $CaCl_2$; [10043-52-4] (3) Magnesium chloride; $MgCl_2$; [7786-30-3] (4) Phosphoric acid, monoammonium salt; $NH_4H_2PO_4$; [7722076-1] (5) Potassium chloride; KCl ; [7447-14-5] (6) Sodium chloride; $NaCl$; [7647-14-5] (7) Urea; CH_4N_2O ; [57-13-6] (8) Water; H_2O ; [7732-18-5]		ORIGINAL MEASUREMENTS: Bandelin, F. J.; Malesh, W. <i>J. Am. Pharm. Assoc., Sci. Ed.</i> 1959, 48, 177-81.																						
VARIABLES: pH at 37°C	PREPARED BY: R. Piekos																							
EXPERIMENTAL VALUES: Solubility of sulfacetamide in a solution containing $CaCl_2$ 0.143, $MgCl_2$ 0.121, $NH_4H_2PO_4$ 0.300, KCl 1.660, $NaCl$ 2.950 and urea 20 g/dm ³ (synthetic urine, Mosher Vehicle) at 37°C																								
<table><tr><th rowspan="2">Equilibrium pH</th><th colspan="2">Solubility</th></tr><tr><th>mg/100 ml</th><th>mol/dm^{3a}</th></tr><tr><td>4.5</td><td>1500</td><td>0.0699</td></tr><tr><td>5.0</td><td>1950</td><td>0.0909</td></tr><tr><td>5.5</td><td>3150</td><td>0.1468</td></tr><tr><td>5.8</td><td>6000</td><td>0.2797</td></tr><tr><td>6.2</td><td>15,000</td><td>0.6992</td></tr><tr><td>6.6</td><td>50,000</td><td>2.3307</td></tr></table>		Equilibrium pH	Solubility		mg/100 ml	mol/dm ^{3a}	4.5	1500	0.0699	5.0	1950	0.0909	5.5	3150	0.1468	5.8	6000	0.2797	6.2	15,000	0.6992	6.6	50,000	2.3307
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^a Calculated by compiler.																								
AUXILIARY INFORMATION																								
METHOD/APPARATUS/PROCEDURE: Excess acetamide was added to aliquots of synthetic urine solns and 1% H_3PO_4 or 1% $NaOH$ solns were used to adjust the pH to the required value. The solns were agitated for 24 h with addn of acid or base to keep them at the desired pH level until equilibrium in pH and concn was attained. Then the solns were filtered and in aliquots acetamide was assayed spectrophotometrically by the method described by Biamonte and Schneller (1). Before detn the soln was refluxed for 1 h with 5% H_2SO_4 to liberate the free amino compd.	SOURCE AND PURITY OF MATERIALS: Nothing specified.																							
	ESTIMATED ERROR: Soly: average values of 2 detns were given. Temp: not specified. pH: not specified.																							
	REFERENCES: 1. Biamonte, A. R.; Schneller, G. E., <i>J. Am. Pharm. Assoc., Sci. Ed.</i> 1952, 41, 341.																							

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9] (2) Ethanol; C ₂ H ₆ O; [64-17-5] (3) Water; H ₂ O; [7732-18-5]		ORIGINAL MEASUREMENTS: Shkadova, A. I. <i>Farm. Zh. (Kiev)</i> <u>1969</u> , <i>24</i> (3), 39-41.	
VARIABLES: Concentration of ethanol		PREPARED BY: R. Piekos	
EXPERIMENTAL VALUES:			
<u>Concentration of ethanol</u>		<u>Solubility at 20°C</u>	
mole %	weight %	10 ² mol kg ⁻¹	g/100 g ^a
0	0	3.02	0.647
10	22.14	11.12	2.383
20	39.01	26.38	5.652
30	52.31	41.59	8.911
40	63.04	58.60	12.555
50	71.90	59.98	12.851
60	79.33	62.18	13.323
70	85.65	60.03	12.862
80	91.10	42.80	9.170
90	95.83	21.21	4.544
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE: Sulfacetamide was equilibrated with the solvent in a water thermostat at 20±0.1°C. The concn of sulfacetamide was detd by alkalimetric titration.		SOURCE AND PURITY OF MATERIALS: Sulfacetamide was prepd from its Na salt by addn of equivalent quantity of 0.1N HCl. The product was washed with water and dried. The EtOH - water mixts were prepd from abs EtOH (purity and source not specified) and distd water.	
		ESTIMATED ERROR: Soly: not specified. Temp: ±0.1°C (author).	
		REFERENCES:	

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> <u>1971</u> , No. 38, 1342-4.	
(2) Formamide; CH ₃ NO; [75-12-7]			
(3) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of formamide		R. Piekos	
EXPERIMENTAL VALUES:			
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.12 \text{ l/mol,}$			
where L _{H₂O} (0.318 _g /50 ml = 2.975 _g × 10 ⁻² mol dm ⁻³ , compiler) and L _s are solubilities of sulfacetamide in water and in aqueous formamide solutions, respectively, and c _s is the concentration of formamide.			
L _s values were supplied by the author in personal communication and are shown below.			
Concentration of formamide		L _s at 20°C	
mol/l	g/100 ml	10 ² mol dm ⁻³ ^a	
0.200	0.700	3.27	
0.400	0.686	3.20	
0.600	0.776	3.62	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.		The source and purity of sulfacetamide was not specified. Anal reagent grade formamide (source not specified) dried over mol sieve was used. Purity of the water was not specified.	
		ESTIMATED ERROR:	
		Soly: not specified. Temp: <u>±</u> 0.05°C (author).	
		REFERENCES:	

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> <u>1971</u> , No. 38, 1342-4.	
(2) Acetamide; C ₂ H ₅ NO; [60-35-5]			
(3) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of acetamide		R. Piekos	
EXPERIMENTAL VALUES:			
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.15 \text{ 1/mol,}$ <p>where L_{H_2O} (0.218₈ g/50 ml = 2.975₈ × 10⁻² mol dm⁻³, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous acetamide solutions, respectively, and c_s is the concentration of acetamide.</p> <p>L_s values were supplied by the author in personal communication and are shown below.</p>			
Concentration of acetamide		L_s at 20°C	
mol/l	g/100 ml	10 ² mol dm ⁻³ ^a	
0.300	0.676	3.16	
0.600	0.802	3.74	
0.900	0.898	4.20	
1.200	0.998	4.66	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.		The source and purity of sulfacetamide was not specified. Anal reagent grade acetamide (source not specified) dried over mol sieve was used. Purity of the water was not specified.	
		ESTIMATED ERROR:	
		Soly: not specified. Temp: <u>±</u> 0.05°C (author).	
		REFERENCES:	

COMPONENTS:		ORIGINAL MEASUREMENTS:																								
(1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> 1971, No. 38, 1342-4.																								
(2) Ethanethioamide; C ₂ H ₅ NS; [62-55-5]																										
(3) Water; H ₂ O; [7732-18-5]																										
VARIABLES:		PREPARED BY:																								
Concentration of ethanethioamide		R. Piekos																								
EXPERIMENTAL VALUES:																										
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.34 \text{ l/mol},$ <p>where L_{H_2O} (0.318₈ g/50 ml 2.975₈ × 10⁻² mol dm⁻³, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous ethanethioamide solutions, respectively, and c_s is the concentration of ethanethioamide. L_s values were supplied by the author in personal communication and are shown below.</p>																										
<table><tr><td rowspan="2">Concentration of ethanethioamide</td><td colspan="2">L_s at 20°C</td></tr><tr><td>mol/l</td><td>10² mol dm⁻³^a</td></tr><tr><td></td><td>g/100 ml</td><td></td></tr><tr><td>0.050</td><td>0.672</td><td>3.14</td></tr><tr><td>0.150</td><td>0.694</td><td>3.24</td></tr><tr><td>0.200</td><td>0.752</td><td>3.51</td></tr><tr><td>0.400</td><td>0.886</td><td>4.14</td></tr><tr><td>0.600</td><td>1.030</td><td>4.81</td></tr></table>				Concentration of ethanethioamide	L_s at 20°C		mol/l	10 ² mol dm ⁻³ ^a		g/100 ml		0.050	0.672	3.14	0.150	0.694	3.24	0.200	0.752	3.51	0.400	0.886	4.14	0.600	1.030	4.81
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^a Calculated by compiler																										
AUXILIARY INFORMATION																										
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:																								
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.		The source and purity of sulfacetamide and water was not specified. Anal reagent grade ethanethioamide (source not specified) dried over mol sieve was used.																								
		ESTIMATED ERROR:																								
		Soly: Not specified. Temp: ±0.05°C (author).																								
		REFERENCES:																								

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> <u>1971</u> , No. 38, 1342-4.	
(2) Propanamide; C ₃ H ₇ NO; [79-05-0]			
(3) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of propanamide		R. Piekos	
EXPERIMENTAL VALUES:			
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.26 \text{ l/mol},$ <p>where L_{H_2O} ($0.318_8 \text{ g/50 ml} = 2.975_8 \times 10^{-2} \text{ mol dm}^{-3}$, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous propanamide solutions, respectively, and c_s is the concentration of propanamide.</p> <p>L_s values were supplied by the author in personal communication and are shown below.</p>			
Concentration of propanamide mol/l		L_s at 20°C	
		g/100 ml	$10^2 \text{ mol dm}^{-3}{}^a$
0.200		0.694	2.24
0.400		0.772	3.60
0.600		0.870	4.06
0.800		0.898	4.20
1.000		1.064	4.97
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.		The source and purity of sulfacetamide and water was not specified. Anal reagent grade propanamide (source not specified) dried over mol sieve was used.	
		ESTIMATED ERROR:	
		Soly: not specified. Temp: $\pm 0.05^\circ\text{C}$ (author).	
		REFERENCES:	

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> <u>1971</u> , No. 38, 1342-4.	
(2) Butanamide; C ₄ H ₉ NO; [541-35-5]			
(3) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of butanamide		R. Piekos	
EXPERIMENTAL VALUES:			
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.30 \text{ l/mol,}$ <p>where L_{H_2O} (0.318 g/50 ml = $2.975 \times 10^{-2} \text{ mol dm}^{-3}$, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous butanamide solutions, respectively, and c_s is the concentration of propanamide. L_s values were supplied by the author in personal communication and are shown below.</p>			
Concentration of butanamide		L_s at 20°C	
mol/l	g/100 ml	$10^2 \text{ mol dm}^{-3}^a$	
0.200	0.710	3.31	
0.400	0.798	3.72	
0.600	0.934	3.36	
0.800	1.080	5.04	
1.000	1.204	5.62	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.		The source and purity of sulfacetamide and water was not specified. Anal reagent grade butanamide (source not specified) dried over mol sieve was used.	
		ESTIMATED ERROR:	
		Soly: not specified. Temp: $\pm 0.05^\circ\text{C}$ (author).	
		REFERENCES:	

COMPONENTS:	ORIGINAL MEASUREMENTS:															
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]	Rohdewald, P. <i>Pharm. Ztg.</i> <u>1971</u> , No. 38, 1342-4.															
(2) Formamide, N,N-dimethyl-; C ₃ H ₇ NO; [68-12-2]																
(3) Water; H ₂ O; [7732-18-5]																
VARIABLES:	PREPARED BY:															
Concentration of N,N-dimethylformamide	R. Piekos															
EXPERIMENTAL VALUES:																
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.32 \text{ l/mol,}$ <p>where L_{H_2O} (0.318 g/50 ml = $2.975 \times 10^{-2} \text{ mol dm}^{-3}$, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous N,N-dimethylformamide solutions, respectively, and c_s is the concentration of N,N-dimethylformamide. L_s values were supplied by the author in personal communication and are shown below.</p>																
<table><tr><td>Concentration of N,N-dimethylformamide</td><td colspan="2">L_s at 20°C</td></tr><tr><td>mol/l</td><td>g/100 ml</td><td>$10^2 \text{ mol dm}^{-3}{}^a$</td></tr><tr><td>0.300</td><td>0.828</td><td>3.86</td></tr><tr><td>0.600</td><td>1.004</td><td>4.69</td></tr><tr><td>0.900</td><td>1.312</td><td>6.12</td></tr></table>		Concentration of N,N-dimethylformamide	L_s at 20°C		mol/l	g/100 ml	$10^2 \text{ mol dm}^{-3}{}^a$	0.300	0.828	3.86	0.600	1.004	4.69	0.900	1.312	6.12
Concentration of N,N-dimethylformamide	L_s at 20°C															
mol/l	g/100 ml	$10^2 \text{ mol dm}^{-3}{}^a$														
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^a Calculated by compiler.																
AUXILIARY INFORMATION																
METHOD/APPARATUS/PROCEDURE:	SOURCE AND PURITY OF MATERIALS:															
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.	The source and purity of sulfacetamide and water was not specified. Anal reagent grade N,N-dimethylformamide (source not specified) dried over mol sieve was used.															
	ESTIMATED ERROR:															
	Soly: not specified. Temp: $\pm 0.05^\circ\text{C}$ (author).															
	REFERENCES:															

COMPONENTS:		ORIGINAL MEASUREMENTS:																
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> 1971, No. 38, 1342-4.																
(2) Adetamide, N-methyl-; C ₃ H ₇ NO; [79-16-3]																		
(3) Water; H ₂ O; [7732-18-5]																		
VARIABLES:		PREPARED BY:																
Concentration of N-methylacetamide		R. Piekos																
EXPERIMENTAL VALUES:																		
<div>$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.25 \text{ l/mol,}$<p>where L_{H_2O} (0.318 g/50 ml = 2.975 x 10⁻² mol dm⁻³, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous N-methylacetamide solutions, respectively and c_s is the concentration of N-methylacetamide. L_s values were supplied by the author in personal communication and are shown below.</p><table><tr><th>Concentration of N-methylacetamide</th><th colspan="2">L_s at 20°C</th></tr><tr><th>mol/l</th><th>g/100 ml</th><th>10² mol dm⁻³^a</th></tr><tr><td>0.300</td><td>0.836</td><td>3.90</td></tr><tr><td>0.600</td><td>0.944</td><td>4.41</td></tr><tr><td>0.900</td><td>1.088</td><td>5.08</td></tr></table><p>^a Calculated by compiler.</p></div>				Concentration of N-methylacetamide	L_s at 20°C		mol/l	g/100 ml	10 ² mol dm ⁻³ ^a	0.300	0.836	3.90	0.600	0.944	4.41	0.900	1.088	5.08
Concentration of N-methylacetamide	L_s at 20°C																	
mol/l	g/100 ml	10 ² mol dm ⁻³ ^a																
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AUXILIARY INFORMATION																		
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:																
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.		The source and purity of sulfacetamide and water was not specified. Anal reagent grade N-methylacetamide (source not specified) dried over mol sieve was used.																
		ESTIMATED ERROR:																
		Soly: not specified. Temp: ±0.05°C (author).																
		REFERENCES:																

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetanide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> 1971, No. 38, 1-2 4.	
(2) Acetamide, N,N-dimethyl-; C ₄ H ₉ NO; [127-19-5]			
(3) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of N,N-dimethylacetamide		R. Piekos	
EXPERIMENTAL VALUES:			
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.41 \text{ l/mol,}$ <p>where L_{H_2O} (0.318 g/50 ml = $2.975 \times 10^{-2} \text{ mol dm}^{-3}$, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous N,N-dimethylacetamide solutions, respectively, and c_s is the concentration of N,N-dimethylacetamide. L_s values were supplied by the author in personal communication and are shown below.</p>			
Concentration of N,N-dimethylacetamide		L_s at 20°C	
mol/l	g/100 ml	$10^2 \text{ mol dm}^{-3}{}^a$	
0.400	0.926	4.32	
0.500	0.760	3.55	
0.600	1.080	5.04	
0.800	1.254	5.85	
1.000	1.500	7.00	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given		The source and purity of sulfacetamide and water was not specified. Anal reagent grade N,N-dimethylacetamide (source not specified) dried over mol sieve was used.	
		ESTIMATED ERROR:	
		Soly: not specified. Temp: $\pm 0.05^\circ\text{C}$ (author).	
		REFERENCES:	

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> 1971, No. 38, 1342-4.	
(2) 3-Pyridinecarboxamide; $C_6H_6N_2O$; [98-92-0]			
(3) Water; H_2O ; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of 3-pyridine-carboxamide		R. Piekos	
EXPERIMENTAL VALUES:			
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.88 \text{ l/mol,}$			
where L_{H_2O} (0.318 g/50 ml = $2.975 \times 10^{-2} \text{ mol dm}^{-3}$, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous 3-pyridinecarboxamide solution, respectively, and c_s is the concentration of 3-pyridinecarboxamide. L_s values were supplied by the author in personal communication and are shown below.			
Concentration of 3-pyridinecarboxamide		L_s at 20°C	
mol/l	g/100 ml	$10^2 \text{ mol dm}^{-3}{}^a$	
0.100	0.854	3.99	
0.150	0.948	4.42	
0.200	1.000	4.67	
0.400	1.424	6.65	
0.600	1.860	8.68	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.		The source and purity of sulfacetamide and water was not specified. Anal reagent grade 3-pyridinecarboxamide (source not specified) dried over mol sieve was used.	
		ESTIMATED ERROR:	
		Soly: not specified.	
		Temp: $\pm 0.05^\circ\text{C}$ (author).	
		REFERENCES:	

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Rohdewald, P. <i>Pharm. Ztg.</i> <u>1971</u> , No. 38, 1342-4.	
(2) 3-Pyridinecarboxamide, N,N-diethyl- (nicetamide) C ₁₀ H ₁₄ N ₂ O; [59-26-7]			
(3) Water; H ₂ O; [7732-18-5]			
VARIABLES:		PREPARED BY:	
Concentration of nicetamide		R. Piekos	
EXPERIMENTAL VALUES:			
$-k_s = \log \frac{L_{H_2O}}{L_s c_s} = 0.94 \text{ l/mol,}$ <p>where L_{H_2O} (0.318 g/50 ml = $2.975 \times 10^{-2} \text{ mol dm}^{-3}$, compiler) and L_s are solubilities of sulfacetamide in water and in aqueous nicetamide solutions, respectively, and c_s is the concentration of nicetamide. L_s values were supplied by the author in personal communication and are shown below.</p>			
Concentration of nicetamide		L_s at 20°C	
mol/l	g/100 ml	$10^2 \text{ mol dm}^{-3}{}^a$	
0.400	1.496	6.982	
0.600	2.108	9.838	
0.800	3.086	14.40	
1.000	4.278	20.00	
^a Calculated by compiler.			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
The solns were equilibrated by agitation for 2 h at 20°C and the sulfacetamide was assayed by differential gravimetric analysis. No details were given.		The source and purity of sulfacetamide and water was not specified. Anal reagent grade nicetamide (source not specified) dried over mol sieve was used.	
		ESTIMATED ERROR:	
		Soly: not specified. Temp: $\pm 0.05^\circ\text{C}$ (author).	
		REFERENCES:	

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Poly(oxy-1,2-ethanediyl), α -hydro- ω -hydroxy (PEG 400); $(C_2H_4O)_n H_2O$; [25322-68-3] 400 (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Khawam, M. N.; Yousef, R. T.; Czetach-Lindenwald, H. <i>Sci. Pharm.</i> <u>1966</u> , <i>34</i> , 209-13.																
VARIABLES: Concentration of PEG 400	PREPARED BY: R. Piekos																
EXPERIMENTAL VALUES: <div data-bbox="420 564 993 1008"> <table border="1"> <caption>Data points estimated from the graph</caption> <thead> <tr> <th>% PEG 400</th> <th>Solubility at 35°C, g/100 g</th> </tr> </thead> <tbody> <tr><td>0</td><td>0</td></tr> <tr><td>20</td><td>4</td></tr> <tr><td>40</td><td>8</td></tr> <tr><td>60</td><td>20</td></tr> <tr><td>80</td><td>30</td></tr> <tr><td>90</td><td>32</td></tr> <tr><td>100</td><td>28</td></tr> </tbody> </table> </div>		% PEG 400	Solubility at 35°C, g/100 g	0	0	20	4	40	8	60	20	80	30	90	32	100	28
% PEG 400	Solubility at 35°C, g/100 g																
0	0																
20	4																
40	8																
60	20																
80	30																
90	32																
100	28																
AUXILIARY INFORMATION																	
METHOD/APPARATUS/PROCEDURE: An earlier described method was employed (1) whereby a 100-ml conical flask contg PEG 400 soln was placed in a drying cabinet at 35°C and an excess of sulfacetamide was added under stirring for 1 h. After 12 h the soln was filtered and decanted and the solute was assayed in the filtrate spectrophotometrically using a Unicam SP 500 spectrophotometer and 1-ml quartz cuvetts. Results were taken from a calibration graph.	SOURCE AND PURITY OF MATERIALS: Neither source nor purity of sulfacetamide and water was specified. PEG 400 was a product of Farbwerke Hoechst (purity not specified). ESTIMATED ERROR: Nothing specified. REFERENCES: 1. Khawam, M. N.; Tawashi, R.; Czetsch-Lindenwald, H. v. <i>Sci. Pharm.</i> <u>1965</u> , <i>33</i> , 90.																

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (Sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Poly(oxy-1,2-ethanediyl), α -hydro- ω -hydroxy- (PEG 600); $(C_2H_4O)_nH_2O$; [25322-68-3] 600 (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Gusyakov, V. P.; Likhol'ot, N. M.; Kutna, I. M. <i>Farm. Zh. (Kiev)</i> <u>1968</u> , 23(6) 56-61.
VARIABLES: One temperature: 21-25°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: <p>Solubility of sulfacetamide in a 5% (by weight) aqueous α-hydro-ω-hydroxy-poly(oxy-1,2-ethanediyl) 600 at room temperature (21-25°C) is 0.852 g/100 ml (3.98×10^{-2} mol dm⁻³, compiler).</p>	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: <p>A small excess of sulfacetamide was added to a 5% (by wt) aq PEG 600 soln, the mixture was sealed in an ampul and agitated for 24 h (1). The concn of sulfacetamide was detd colorimetrically (2).</p>	SOURCE AND PURITY OF MATERIALS: <p>Sulfacetamide: neither source nor purity was specified. PEG 600 was of the Austrian or West German origin. Its purity was not specified. Purity of the water was not specified.</p>
	ESTIMATED ERROR: <p>Nothing specified.</p>
	REFERENCES: 1. Gusyakov, V. P.; Likhol'ot, N. M.; Kuta, I. M. <i>Farm. Zh. (Kiev)</i> <u>1967</u> , 22(3), 34. 2. Predchetenskiĭ, B. E.; Borovskaya, V. M.; Morgolina, L. T. <i>Laboratornye metody issledovaniya, Medgiz, Moscow</i> <u>1950</u> , p. 371.

COMPONENTS: (1) Acetamide, N-[4-aminophenyl]-sulfonyl]- (sulfacetamide; $C_8H_{10}N_2O_3S$; [144-80-9] (3) Poly(oxy-1,2-ethanediyl), α -hydro- ω -hydroxy- (PEG 4000); $C_2H_4O)_nH_2O$; [25322-68-3] 4000 (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Khawam, M. N.; Yousef, R. T.; Czetsch-Lindenwald, H. <i>Sci. Pharm.</i> <u>1966</u> , 34, 209-13.														
VARIABLES: Concentration of PEG 4000	PREPARED BY: R. Piekos														
EXPERIMENTAL VALUES: <div data-bbox="454 574 1035 1126" data-label="Figure"> <table border="1"> <caption>Data points estimated from the graph</caption> <thead> <tr> <th>% PEG 4000</th> <th>Solubility at 35°C, g/100 g</th> </tr> </thead> <tbody> <tr><td>5</td><td>1.5</td></tr> <tr><td>10</td><td>2.0</td></tr> <tr><td>15</td><td>2.5</td></tr> <tr><td>20</td><td>3.2</td></tr> <tr><td>25</td><td>4.1</td></tr> <tr><td>30</td><td>5.8</td></tr> </tbody> </table> </div>		% PEG 4000	Solubility at 35°C, g/100 g	5	1.5	10	2.0	15	2.5	20	3.2	25	4.1	30	5.8
% PEG 4000	Solubility at 35°C, g/100 g														
5	1.5														
10	2.0														
15	2.5														
20	3.2														
25	4.1														
30	5.8														
AUXILIARY INFORMATION															
METHOD/APPARATUS/PROCEDURE: An earlier described method was employed (1) whereby a 100-ml conical flask contg a PEG 4000 soln was placed in a drying cabinet at 35°C and an excess of sulfacetamide was added under stirring for 1 h. After 12 h the soln was filtered or decanted and the solute was assayed in the filtrate spectrophotometrically using a Unicam SP 500 spectrophotometer and 1-ml quartz cuvetts. Results were taken from a calibration graph.	SOURCE AND PURITY OF MATERIALS: Neither source nor purity of sulfacetamide and water were specified. PEG 4000 was a product of Farbwerke Hoechst (purity not specified). ESTIMATED ERROR: Nothing specified. REFERENCES: 1. Khawam, M. N.; Tawashi, R.; Czetsch-Lindenwald, H. v. <i>Sci. Pharm.</i> <u>1965</u> , 33, 90.														

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Poly(oxy-1,2-ethanediyl), α -hydro- ω -hydroxy- (PEG 4000); $(C_2H_4O)_nH_2O$; [25322-68-3] 4000 (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Gussyakov, V. P.; Likholt'ot, N. M.; Kutna, I.M. <i>Farm. Zh. (Kiev)</i> 1968, 23(6), 56-61.
VARIABLES: One temperature: 21-25°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in a 5% (by weight) aqueous α -hydro- ω -hydroxy-poly(oxy-1,2-ethanediyl) 4000 at room temperature (21-25°C is 0.852 g/100 ml ($3.98 \times 10^{-2} \text{ dm}^{-3}$, compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: A small excess of sulfacetamide was added to a 5% (by wt) aq PEG 4000 soln, the mixture was sealed in an ampul and agitated for 25 h (1). The concn of sulfacetamide was detd colorimetrically (2).	SOURCE AND PURITY OF MATERIALS: Sulfacetamide: neither source nor purity was specified. PEG 4000 was of the Austrian or West German origin. Its purity was not specified. Purity of the water was not specified. ESTIMATED ERROR: Nothing specified. REFERENCES: 1. Gussyakov, V. P.; Likholt'ot, N. M.; Kutna, I. M. <i>Farm. Zh. (Kiev)</i> 1967, 22(3), 34. 2. Predchetenskii, B. E.; Borovskaya, V. M.; Morgolina, L. T. <i>Laboratornye metody issledovaniya, Medgiz, Moscow</i> 1950, p. 371.

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl] (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Sorbitan monolaurate, polyoxyethylene derivatives (Tween 20); [9005-64-5] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Khawam, M. N.; Yousef, R. T.; Czetsch- Lindenwald, H. <i>Sci. Pharm.</i> <u>1966</u> , <i>34</i> , 209-13.																
VARIABLES: Concentration of Tween 20	PREPARED BY: R. Piekos																
EXPERIMENTAL VALUES: <div data-bbox="454 504 1000 1179" data-label="Figure"> <table border="1"> <caption>Data points from the solubility graph</caption> <thead> <tr> <th>% Tween 20</th> <th>Solubility at 35°C, g/100 g</th> </tr> </thead> <tbody> <tr><td>10</td><td>2.5</td></tr> <tr><td>20</td><td>4</td></tr> <tr><td>40</td><td>8</td></tr> <tr><td>60</td><td>16</td></tr> <tr><td>80</td><td>28</td></tr> <tr><td>90</td><td>31</td></tr> <tr><td>100</td><td>29</td></tr> </tbody> </table> </div>		% Tween 20	Solubility at 35°C, g/100 g	10	2.5	20	4	40	8	60	16	80	28	90	31	100	29
% Tween 20	Solubility at 35°C, g/100 g																
10	2.5																
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AUXILIARY INFORMATION																	
METHOD/APPARATUS/PROCEDURE: An earlier described method was employed (1) whereby a 100-ml conical flask contg a Tween 20 soln was placed in a drying cabinet at 35°C and an excess of sulfacetamide was added under stirring for 1 h. After 12 h the soln was filtered or decanted and the solute was assayed in the filtrate spectrophotometrically using a Unicam SP 500 spectrophotometer and 1-ml quartz cuvetts. Results were taken from a calibration graph.	SOURCE AND PURITY OF MATERIALS: Neither source nor purity of sulfacetamide and water were reported. Tween 20 was supplied by Atlas-Goldschmidt A. G., Essen (purity not specified). ESTIMATED ERROR: Nothing specified. REFERENCES: 1. Khawam, M. N.; Tawashi, R.; Czetsch-Lindenwald, H. v. <i>Sci. Pharm.</i> <u>1965</u> , <i>33</i> , 90.																

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Sorbitan monolaurate, polyoxyethylene derivatives (Tween 20); [9005-64-5] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Gussyakov, V. P.; Likholt'ot, N. M.; Kutna, I. M. <i>Farm. Zh. (Kiev)</i> <u>1967</u> , 22(3) 34-9.
VARIABLES: One temperature: 20°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: $S/S_o = 1.6 \text{ at } 20^\circ\text{C}$ <p>where S is the solubility of sulfacetamide in a 2% by weight aqueous Tween 20 solution, and</p> <p>S_o is the solubility of sulfacetamide in water (0.50 g/100 ml).</p> <p>Hence $S = 0.80 \text{ g/100 ml}$ ($3.7 \times 10^{-2} \text{ mol dm}^{-3}$) - compiler.</p>	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: An excess of sulfacetamide in a 2% by wt aq Tween 20 soln was equilibrated for 24 h in an ampul immersed in a water thermostat. Aliquots of the satd soln were withdrawn through a filter and the sulfacetamide content was assayed in the filtrate photometrically.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide conformed to the requirements of the State Pharmacopeia IX. Tween 20 was a product of Gee Lawson, England. Purity of the water was not specified.
	ESTIMATED ERROR: Soly: not specified. Temp: $\pm 0.1^\circ\text{C}$ (authors).
	REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl) sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Sorbitan monopalmitate, polyoxyethylene derivatives (Tween 40); [9005-66-7] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Gusyakov, V. P.; Likholt', N. M.; Kutna, I. M. <i>Farm. Zh. (Kiev)</i> 1967 , <i>22</i> (3), 34-9.
VARIABLES: One temperature: 20°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: $S/S_o = 1.72 \text{ at } 20^\circ\text{C}$ where S is the solubility of sulfacetamide in a 2% by weight aqueous Tween 40 solution, and S_o is the solubility of sulfacetamide in water (0.50 g/100 ml). Hence $S = 0.86 \text{ g/100 ml}$ ($4.0 \times 10^{-2} \text{ mol dm}^{-3}$) - compiler.	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: An excess of sulfacetamide in a 2% by wt aq Tween 40 soln was equilibrated for 24 h in an ampul immersed in a water thermostat. Aliquots of the satd soln were withdrawn through a filter and the sulfacetamide content was assayed in the filtrate photometrically.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide conformed to the requirements of the State Pharmacopeia IX. Tween 40 was a product of Gee Lawson, England. Purity of the water was not specified. ESTIMATED ERROR: Soly: not specified. Temp: $\pm 0.1^\circ\text{C}$ (authors). REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-8] (2) Sorbitan monooleate, polyoxyethylene derivatives (Tween 80); [9005-65-6] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Gusyakov, V. P.; Likhol'ot, N. M.; Kutna, I. M. <i>Farm. Zh. (Kiev)</i> 1967 , <i>22</i> (3), 34-9.
VARIABLES: One temperature: 20°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: $S/S_o = 1.7 \text{ at } 20^\circ\text{C}$ where S is the solubility of sulfacetamide in a 2% by weight aqueous Tween 80 solution, and S_o is the solubility of sulfacetamide in water (0.50 g/100 ml). Hence $S = 0.85 \text{ g/100 ml}$ ($4.0 \times 10^{-2} \text{ mol dm}^{-3}$) - compiler.	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: An excess of sulfacetamide in a 2% by wt aq Twen 80 soln was equilibrated for 24 h in an ampul immersed in a water thermostat. Aliquots of the satd soln were withdrawn through a filter and the sulfacetamide content was assayed in the filtrate photo-metrically.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide conformed to the requirements of the State Pharmacopeia IX. Tween 80 was a product of Gee Lawson, England. Purity of the water was not specified.
	ESTIMATED ERROR: Soly: not specified. Temp: $\pm 0.1^\circ\text{C}$ (authors).
	REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) White petrolatum (liquid petrolatum)	ORIGINAL MEASUREMENTS: Whitworth, C. W.; Becker, C. H. <i>J. Pharm. Sci.</i> <u>1965</u> , <i>54</i> (4), 569-73.
VARIABLES: One temperature: 37.5°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: <p>Solubility of sulfacetamide in white petrolatum (liquid petrolatum) at 37.5°C is 0.089 mg% (4.1×10^{-6} mol dm⁻³ solution, compiler).</p>	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: A satd soln of sulfacetamide in liquid petrolatum was made and filtered carefully at a const temp to remove suspended particles. A portion of the soln was shaken for 4 h with 10 ml of EtOH. The alcoholic layer was centrifuged for 30 min. Aliquot portions of the alcoholic solns were allowed to evap to dryness, a trichloroacetic acid soln added, and subsequently the Marshall reagents. From the intensity of the color developed it was impossible to det the amt of the drug extd by the process utilized. A Klett-Summerson colorimeter with a No 54 filter was employed to det the color intensity, which was compared to that of standard solns.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide (N.F. grade) was from Ruger Chemical Co., Inc. White petrolatum (liquid petrolatum) (U.S.P. grade) was from Fisher Scientific Co. ESTIMATED ERROR: Soly: not specified. Temp: $\pm 1^\circ\text{C}$ (authors). REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9] (2) Sorbitan, (Z)-9-octadecenoate (2:3) (Arlacel 83); [8007-43-0] (3) White petrolatum (liquid petrolatum)	ORIGINAL MEASUREMENTS: Whitworth, C. W.; Becker, C. H. <i>J. Pharm. Sci.</i> <u>1965</u> , <i>54</i> (4), 569-73														
VARIABLES: Concentration of Arlacel 83	PREPARED BY: R. Piekos														
EXPERIMENTAL VALUES: <table><tr><th rowspan="2">Concentration of Arlacel 83 %</th><th colspan="2">Solubility at 37.5°C</th></tr><tr><th>mg%</th><th>10⁵ mol dm⁻³ soln^a</th></tr><tr><td>1</td><td>0.150</td><td>0.700</td></tr><tr><td>5</td><td>0.906</td><td>4.22</td></tr><tr><td>10</td><td>1.761</td><td>8.21</td></tr></table> <p>^a Calculated by compiler.</p>		Concentration of Arlacel 83 %	Solubility at 37.5°C		mg%	10 ⁵ mol dm ⁻³ soln ^a	1	0.150	0.700	5	0.906	4.22	10	1.761	8.21
Concentration of Arlacel 83 %	Solubility at 37.5°C														
	mg%	10 ⁵ mol dm ⁻³ soln ^a													
1	0.150	0.700													
5	0.906	4.22													
10	1.761	8.21													
AUXILIARY INFORMATION															
METHOD/APPARATUS/PROCEDURE: <p>A satd soln of sulfacetamide in the solvent was made and filtered carefully at a const temp to remove all suspended particles. A 5-ml portion of the soln was shaken for 4 h with 100 ml of EtOH. The alcoholic layer was centrifuged for 30 min. Aliquot portions of the alcoholic solns were allowed to evap to dryness, a trichloroacetic acid soln was added, and subsequently the Marshall reagents. From the intensity of the color developed it was possible to det the amt of the drug extd by the process utilized. A Klett-Summerson colorimeter with a No 54 filter was employed to det the color intensity, which was compared to that of standard solns.</p>	SOURCE AND PURITY OF MATERIALS: <p>Sulfacetamide (N.F. grade) was from Ruger Chemical Co, Inc. Arlacel 83 (Lot No 129) was from Atlas Powder Co. (Purity not specified). White petrolatum (U.S.P. grade) was from Fisher Scientific Co.</p> ESTIMATED ERROR: <p>Soly: not specified. Temp: $\pm 1^{\circ}\text{C}$ (authors).</p> REFERENCES:														

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide) $C_8H_{10}N_2O_3S$; [144-80-9] (2) Cottonseed oil	ORIGINAL MEASUREMENTS: Whitworth, C. W.; Becker, C. H. <i>J. Pharm. Sci.</i> <u>1965</u> , <i>54</i> (4), 569-73.
VARIABLES: One temperature: 37.5°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in cottonseed oil at 37.5°C is 4.734 mg% (2.212 x 10 ⁻⁴ mol dm ⁻³ solution, compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: A satd soln of sulfacetamide in cottonseed oil was made and filtered carefully at a const temp to remove suspended particles. A portion of the soln was shaken for 4 h with 100 ml of EtOH. The alcoholic layer was centrifuged for 30 min. Aliquot portions of the alcoholic solns were allowed to evap to dryness, a trichloroacetic acid soln added, and subsequently the Marshall reagents. From the intensity of the color developed it was possible to det the amt of the drug extd by the process utilized. A Klett-Summerson colorimeter with a No 54 filter was employed to det the color intensity, which was compared to that of standard solns.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide (N.F. grade) was from Ruger Chemical Co., Inc. Neither source nor purity of the cottonseed oil was specified. ESTIMATED ERROR: Soly: not specified. Temp: $\pm 1^\circ\text{C}$ (authors). REFERENCES:

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]		Whitworth, C. W.; Becker, C. H. J. Pharm. Sci. 1965, 54(4), 569-73.	
(2) Cottonseed oil			
(3) Sorbitan, (Z)-9-octadecenoate (2:3) (Arlacel 83); [8007-43-0]			
VARIABLES:		PREPARED BY:	
Concentration of Arlacel 83		R. Piekos	
EXPERIMENTAL VALUES:			
Concentration of Arlacel 83		Solubility at 37.5°C	
%	mg%	10 ⁴ mol dm ⁻³ soln ^a	
1	5.675	2.648	
5	6.950	3.244	
10	8.45	3.94	
^a Calculated by compiler			
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
A satd soln of sulfacetamide in the solvent was made and filtered carefully at a const temp to remove all suspended matter. A 5-ml portion of the soln was shaken for 4 h with 100 ml of EtOH. The alcoholic layer was centrifuged for 30 min. Aliquot portions of the alcoholic solns were allowed to evap to dryness, a trichloroacetic acid soln was added, and subsequently the Marshall reagents. From the intensity of the color developed it was possible to det the amt of the drug extd by the process utilized. A Klett-Summerson colorimeter with a No 54 filter was employed to det the color intensity, which was compared to that of standard solns.		Sulfacetamide (N.F. grade) was from Ruger Chemical Co., Inc. Neither source nor purity of the cottonseed oil was specified. Arlacel 83 (Lot No 129) was from Atlas Powder Co. (purity not specified).	
		ESTIMATED ERROR:	
		Soly: not specified. Temp: <u>±1</u> °C (authors).	
		REFERENCES:	

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Methane, trichloro- (chloroform); $CHCl_3$; [67-66-3]	ORIGINAL MEASUREMENTS: Yamazaki, M.; Aoki, M.; Kamada, A.; Yata, N. <i>Yakuzaiigaku</i> <u>1967</u> , 27(1), 37-40.
VARIABLES: One temperature: 30°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: <p>Solubility of sulfacetamide in chloroform at 30°C is 3.60 mmol/L (0.77 g dm⁻³, compiler).</p>	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: <p>Sulfacetamide (0.5 g) was placed in an L-shaped tube together with 20 ml of chloroform. The mixt was shaken in a thermostat until equilibrium was attained. The sulfacetamide was then assayed in the supernatant spectrophotometrically at 545 nm on a Beckmann DU spectrophotometer. The results were taken from a calibration graph.</p>	SOURCE AND PURITY OF MATERIALS: <p>Nothing specified.</p>
	ESTIMATED ERROR: Soly: not specified. Temp: $\pm 1^\circ C$ (authors).
	REFERENCES:

COMPONENTS:				ORIGINAL MEASUREMENTS:			
(1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); C ₈ H ₁₀ N ₂ O ₃ S; [144-80-9]				Gutierrez, F. H. Anales fis. quim. Madrid) 1945, 41, 537-60.			
(2) 2-Propanone (acetone); C ₃ H ₆ O; [67-64-1]							
VARIABLES:				PREPARED BY:			
Temperature				R. Piekos			
EXPERIMENTAL VALUES:							
t/°C	G ^a	E ^b	X _g /l ^c	mol/l ^d acetone	mmol/mol acetone	1:X _g ^e	1 + X _{cc} ^f
0	13.858	12.085	112.887	527	37	7.22	8.86
5	15.005	13.047	121.360	566	41	6.66	8.02
10	16.450	14.126	132.094	617	45	6.08	7.57
15	18.408	14.546	146.730	685	50	5.43	6.82
20	20.948	17.328	165.741	774	57	4.77	6.03
25	23.904	19.336	187.694	876	65	4.19	5.33
30	27.751	21.723	216.265	1095	75	3.50	4.62
40	38.144	27.611	292.717	1361	103	2.62	3.42
45	45.913	31.465	349.582	1632	124	2.18	2.86
50	59.893	37.458	451.592	2127	163	1.67	2.21
<p>^a $G = \frac{p \ 100}{P - p}$, where p and P are the weights of solute and solution, resp.</p> <p>^b $E = \frac{G \ 100}{G + 100}$; ^c g/l acetone; ^d should be mmol/l acetone (compiler);</p> <p>^e g of acetone required to dissolve 1 g of solute;</p> <p>^f volume (cm³) of acetone required to dissolve 1 g of solute.</p>							
AUXILIARY INFORMATION							
METHOD/APPARATUS/PROCEDURE:				SOURCE AND PURITY OF MATERIALS:			
A special all-glass app was constructed enabling the prepn of satd solns, agitation by bubbling a stream of acetone-satd N, filtration, and distn off the solvent without contact with air. Two exchange-able dissoln vessels of 15 and 8 cm ³ working capacity were used depending on the soly of solute. The app was immersed in a thermostat. The vols of acetone used were 15 or 5 cm ³ , and the equilibration time was 2-2.5 h. The satd solns were filtered, weighed, the solvent was distd off, the residues were dried at 105°C, weighed, and examd for the presence of solvated acetone.				The source of the materials was not specified. Pure, anhydrous acetone was used. The absence of impurities and water was confirmed by procedures of the German Pharmacopeia VI and Spanish Pharmacopeia VIII. The purity of sulfacetamide was not specified.			
				ESTIMATED ERROR:			
				Soly: measurements were repeated until 2 values not differing in the second decimal were obtained (author). Temp: +0.1°C (author).			
				REFERENCES:			

COMPONENTS: (1) Acetamide, N-[(4-aminophenyl)-sulfonyl]- (sulfacetamide); $C_8H_{10}N_2O_3S$; [144-80-9] (2) Poly(oxy-1,2-ethanediyl), α -hydro- ω -hydroxy- (PEG 400); $(C_2H_4O)_nH_2O$; [25322-68-3]	ORIGINAL MEASUREMENTS: Gusyakov, V. P.; Likhol'ot, N. M.; Kutna, I. M. <i>Farm. Zh. (Kiev)</i> <u>1968</u> , <i>23</i> (6), 56-61.
VARIABLES: One temperature: 21-25°C	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of sulfacetamide in α -hydro- ω -hydroxypoly(oxy-1,2-ethanediyl) 400 at room temperature (21-25°C) is 19.9% by weight (1.16 mol kg ⁻¹ PEG 400, compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: Small quantities (2-4 mg), of sulfaceta- mide were added to a known quantity of PEG 400 under stirring until satn was attained.	SOURCE AND PURITY OF MATERIALS: Sulfacetamide: neither source nor purity was specified. PEG 400: source not specified; sp gr 1.127 g cm ⁻³ ; temp of solidification approx 6°C; refractive index 1.466 (temp not indicated). ESTIMATED ERROR: REFERENCES:

COMPONENTS: (1) Acetamide, N-[(4-acetylamino)-phenyl] sulfonyl - (acetyl sulfacetamide); $C_{10}H_{12}N_2O_4S$; [5626-90-4] (2) Phosphoric acid, disodium salt; Na_2HPO_4 ; [7558-94-4] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Krüger-Thiemer, E. <i>Arch. Dermatol. Syphilis</i> <u>1942</u> , 183, 90-116.
VARIABLES: One temperature: ca 20°C; one pH: 8.74	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of acetyl sulfacetamide in a 0.705M (10%) Na_2HPO_4 solution of pH 8.74 at room temperature (about 20°C) is 2.040 g% (7.959×10^{-2} mol dm ⁻³ solution, compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: Acetyl sulfacetamide (0.5 g) was dissolved in the 0.705M (10%) Na_2HPO_4 soln of pH 8.74, shaken for 2 h at room temp (about 20°C), and filtered. The filtrate was treated with equal vol of 2N HCl and refluxed for 15 min. After proper diln, a 1-cm ³ aliquot was withdrawn, acidified, cooled, and the sulfonamide content was detd colorimetrically by the Marshall method modified by Kimmig (1) using an Authenreith colorimeter. The pH value was detd on an ultraionograph using a glass electrode.	SOURCE AND PURITY OF MATERIALS: Acetyl sulfacetamide (source not specified) gave no coloration upon diazotization of its satd soln, thus showing absence of sulfanilamide. The source and purity of the remaining materials was not specified. ESTIMATED ERROR: Soly: precision $\pm 5\%$ (author). Temp: not specified. pH: ± 0.05 pH unit (author). REFERENCES: 1. Kimmig, J. <i>Arch. Dermatol.</i> <u>1938</u> , 176, 722; <i>Erg. Hyg.</i> <u>1941</u> , 24, 398.

COMPONENTS: (1) Acetamide, N-[[(4-acetylamino) phenyl]sulfonyl] - (acetyl sulfacetamide); $C_{10}H_{12}N_2O_4S$; [5626-90-4] (2) Phosphoric acid, monopotassium salt; KH_2PO_4 ; [7778-77-0] (3) Water; H_2O ; [7732-18-5]	ORIGINAL MEASUREMENTS: Krüger-Thiemer, E. <i>Arch. Dermatol. Syphilis</i> <u>1942</u> , 183, 90-116.
VARIABLES: One temperature: ca 20°C; one pH: 4.37	PREPARED BY: R. Piekos
EXPERIMENTAL VALUES: Solubility of acetyl sulfacetamide in a 0.735 M (10%) KH_2PO_4 solution of pH 4.37 at room temperature (about 20°C) is 0.028 g% (1.1×10^{-3} mol dm ⁻³ solution, compiler).	
AUXILIARY INFORMATION	
METHOD/APPARATUS/PROCEDURE: Acetyl sulfacetamide (0.5 g) was dissolved in the 0.735 M (10% KH_2PO_4 soln, shaken for 2 h at room temp (about 20°C), and filtered. The filtrate was treated with equal vol of 2N HCl, and refluxed for 15 min. After proper diln, a 1-cm ³ aliquot was withdrawn, acidified, cooled, and the sulfonamide content was detd colorimetrically by the Marshall method modified by Kimmig (1) using an Authenreith colorimeter. The pH value was detd on an ultraionograph using a glass electrode.	SOURCE AND PURITY OF MATERIALS: Acetyl sulfacetamide (source not specified) gave no coloration upon diazotization of its satd soln, thus showing absence of sulfanilamide. The source and purity of the remaining materials was not specified. ESTIMATED ERROR: Soly: precision $\pm 5\%$ (author). Temp: not specified. pH: ± 0.05 pH unit (author). REFERENCES: 1. Kimmig, J. <i>Arch. Dermatol.</i> <u>1938</u> , 176, 722; <i>Erg. Hyg.</i> <u>1941</u> , 24, 398.