

<b>COMPONENTS:</b> (1) Benzenesulfonamide, 4-amino-N-(aminothioxomethyl)-, (Badional); $C_7H_9N_3O_2S_2$ ; [515-49-1] (2) Ethanol; $C_2H_6O$ ; [64-17-5] (3) Water; $H_2O$ ; [7732-18-5]	<b>ORIGINAL MEASUREMENTS:</b> Kuhnert-Brandstätter, M.; Burger, A., <i>Pharm. Ind.</i> 1972, 34, 353-6.																		
<b>VARIABLES:</b> Temperature	<b>PREPARED BY:</b> R. Piekos																		
<b>EXPERIMENTAL VALUES:</b>  Saturation solubility, $c_s$ , of forms I and II of Badional in 96% ethanol. Form I (+ + +), form II (•••)  <table border="1"> <caption>Estimated data points from the solubility graph</caption> <thead> <tr> <th>Temperature (°C)</th> <th>Form I Solubility (mg/100 ml)</th> <th>Form II Solubility (mg/100 ml)</th> </tr> </thead> <tbody> <tr> <td>10</td> <td>1250</td> <td>1000</td> </tr> <tr> <td>20</td> <td>1350</td> <td>1300</td> </tr> <tr> <td>30</td> <td>1550</td> <td>1600</td> </tr> <tr> <td>40</td> <td>2000</td> <td>2300</td> </tr> <tr> <td>50</td> <td>3000</td> <td>3600</td> </tr> </tbody> </table>		Temperature (°C)	Form I Solubility (mg/100 ml)	Form II Solubility (mg/100 ml)	10	1250	1000	20	1350	1300	30	1550	1600	40	2000	2300	50	3000	3600
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<b>AUXILIARY INFORMATION</b>																			
<b>METHOD/APPARATUS/PROCEDURE:</b>  An excess of Badional was added to 96% ethanol and stirred until no changes in concn were obsd (2 - 48 h). Aliquots were withdrawn by means of a pipet equipped with a filter, which was preheated to a desired temp, and Badional was assayed spectrophotometrically at 269 nm. The identity of the solid phase was detd by thermomicroscopy and IR spectroscopy.	<b>SOURCE AND PURITY OF MATERIALS:</b>  Both forms were obtained by crystn of a compd prep from 96% EtOH; form I by cooling a hot satd soln to room temp and seeding with crystals of this form, form II by cooling the soln to $\sim 40^\circ C$ and scratching the walls.  Purity of the 96% EtOH was not specified.																		
	<b>ESTIMATED ERROR:</b>  Soly: not specified. Temp: $\pm 0.1^\circ C$ (authors).																		
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<b>COMPONENTS:</b> (1) Benzenesulfonamide, 4-amino-N-(aminothioxomethyl)-, (Badional); $C_7H_9N_3O_2S_2$ ; [515-49-1] (2) Ethanol; $C_2H_6O$ ; [64-17-5] (3) Water; $H_2O$ ; [7732-18-5]		<b>ORIGINAL MEASUREMENTS:</b> Kuhnert-Brandstätter, M.; Burger, A., <i>Pharm. Ind.</i> 1972, 34, 353-6.		
<b>VARIABLES:</b> Temperature		<b>PREPARED BY:</b> R. Piekos		
<b>EXPERIMENTAL VALUES:</b>				
Saturation solubility in 96% ethanol of crystalline form				
$t/^\circ C$	I		II	
	g/100 ml	$mol\ dm^{-3}^a$	g/100 ml	$mol\ dm^{-3}^a$
10.0			1.00	0.0432
10.1	1.06	0.0458		
14.6			1.12	0.0484
15.2	1.18	0.0510		
20.0	1.37	0.0592	1.29	0.0558
25.2			1.54	0.0666
25.4	1.57	0.0679		
30.9			1.81	0.0783
31.0	1.79	0.0774		
37.0			2.21	0.0955
37.1	2.05	0.0886		
41.2	2.30	0.0994	2.52	0.109
46.0	2.68	0.116		
46.2			3.00	0.130
50.8	3.06	0.132		
51.0			3.50	0.151
<sup>a</sup> Calculated by compiler.				
<b>AUXILIARY INFORMATION</b>				
<b>METHOD/APPARATUS/PROCEDURE:</b> An excess of Badional was added to 96% ethanol and stirred until no changes in concn were obsd (2-48 h). Aliquots were withdrawn by means of a pipet equipped with a filter, which was preheated to a desired temp, and Badional was assayed spectrophotometrically at 269 nm. The identity of the solid phase was detd by thermomicroscopy and IR spectroscopy.			<b>SOURCE AND PURITY OF MATERIALS:</b> Both forms of Badional were obtained by crystn of a comprep from 96% EtOH: form I, mp 179-181°C, by cooling a hot satd soln to room temp and seeding with crystals of this form; form II, mp 171°C, by cooling the soln to 40°C and scratching the walls. Purity of the 96% ethanol was not specified.	
			<b>ESTIMATED ERROR:</b> Soly: not specified. Temp: $\pm 0.1^\circ C$ (authors).	
			<b>REFERENCES:</b>	

<b>COMPONENTS:</b> (1) Benzenesulfonamide; 4-amino-N-(aminothioxomethyl)- (sulfathiourea); $C_7H_9N_3O_2S_2$ ; [515-49-1] (2) Ethanol; $C_2H_6O$ ; [64-17-5] (3) 1,2,3-Propanetriol; $C_3H_8O_3$ ; [56-81-5] (4) Water; $H_2O$ ; [7732-18-5]	<b>ORIGINAL MEASUREMENTS:</b> Dolique, R.; Foucault, J. <i>Trav. soc. pharm. Montpellier</i> <u>1952</u> , 12, 145-53.
<b>VARIABLES:</b> One temperature: 26-28°C	<b>PREPARED BY:</b> R. Piekos
<b>EXPERIMENTAL VALUES:</b> <p>Solubility of sulfathiourea in a mixture of 1,2,3-propanetriol and 95° ethanol (2:1 by wt) at 26-28°C is 7.72% (0.362 mol kg<sup>-1</sup> solvent, compiler).</p>	
<b>AUXILIARY INFORMATION</b>	
<b>METHOD/APPARATUS/PROCEDURE:</b> <p>The sulfathiourea content was detd by diazotization of the amine group in a cold acidified 0.1N <math>KNO_2</math> soln. An excess of <math>KNO_2</math> was detected by using iodinated starch.</p>	<b>SOURCE AND PURITY OF MATERIALS:</b> Nothing specified.  <b>ESTIMATED ERROR:</b> Nothing specified.  <b>REFERENCES:</b>

<b>COMPONENTS:</b> (1) Benzenesulfonamide, 4-amino-N-(aminothioxomethyl)- (sulfathiourea); $C_7H_9N_3O_2S_2$ ; [515-49-1] (2) Ethanol; $C_2H_6O$ ; [64-17-5] (3) 1,2,3-Propanetriol; $C_3H_8O_3$ ; [56-81-5] (4) Urea; $CH_4N_2O$ ; [57-13-6] (5) Water; $H_2O$ ; [7732-18-5]	<b>ORIGINAL MEASUREMENTS:</b> Doliq, R.; Foucault, J. <i>Trav. soc. pharm. Montpellier</i> <u>1952</u> , <i>12</i> , 145-53.
<b>VARIABLES:</b>  One temperature: 26-28°C	<b>PREPARED BY:</b>  R. Piekos
<b>EXPERIMENTAL VALUES:</b>  <p>Solubility of sulfathiourea at 26-28°C in a saturated solution of urea in a mixture of 1,2,3-propanetriol and 95° ethanol (2:1 by wt), containing 54.5 g of urea per 100 g of the mixture, is 9.47% (0.452 mol kg<sup>-1</sup> solvent, compiler).</p>	
<b>AUXILIARY INFORMATION</b>	
<b>METHOD/APPARATUS/PROCEDURE:</b>  The sulfathiourea content was detd by diazotization of the amine group in a cold acidified 0.1N KNO <sub>2</sub> soln. An excess of KNO <sub>2</sub> was detected by using iodinated starch.	<b>SOURCE AND PURITY OF MATERIALS:</b>  Nothing specified.
<b>ESTIMATED ERROR:</b>  Nothing specified.	
<b>REFERENCES:</b>	

<b>COMPONENTS:</b> (1) Benzenesulfonamide, 4-amino-N-[imino-(methylthio)methyl]- (sulfamethylisothiourea); $C_8H_{11}N_3O_2S$ ; [2651-18-5] (2) Water; $H_2O$ ; [7732-18-5]	<b>ORIGINAL MEASUREMENTS:</b> Becher, R.; Leya, S. <i>Experientia</i> <u>1946</u> , 2 459-60.
<b>VARIABLES:</b> One temperature: 18-19°C.	<b>PREPARED BY:</b> R. Piekos
<b>EXPERIMENTAL VALUES:</b> Solubility of sulfamethylisothiourea in water at room temperature (18-19°C) is 19 mg% ( $8.9 \times 10^{-4}$ mol $dm^{-3}$ , compiler).	
<b>AUXILIARY INFORMATION</b>	
<b>METHOD/APPARATUS/PROCEDURE:</b> After standing for more than two days the soln of sulfamethylisothiourea in water was filtered and the sulfonamide was assayed in the filtrate colorimetrically by the method of Druey and Oesterheld (1).	<b>SOURCE AND PURITY OF MATERIALS:</b> Nothing specified.  <b>ESTIMATED ERROR:</b> Nothing specified.  <b>REFERENCES:</b> 1. Druey, J.; Oesterheld, G. <i>Helv. Chim. Acta</i> <u>1942</u> , 25, 753.

<b>COMPONENTS:</b> (1) Benzenesulfonamide, 4-amino-N-[imino-(methylthio)methyl]- (sulfamethylisothiurea); $C_8H_{11}N_3O_2S$ ; [2651-18-5] (2) Sodium chloride; NaCl; [7647-14-5] (3) Water; $H_2O$ ; [7732-18-5]	<b>ORIGINAL MEASUREMENTS:</b> Becher, R.; Leya, S. <i>Experientia</i> <u>1946</u> , <i>2</i> , 459-60.
<b>VARIABLES:</b> One temperature: 18-19°C	<b>PREPARED BY:</b> R. Piekos
<b>EXPERIMENTAL VALUES:</b> <p>Solubility of sulfamethylisothiurea in a 5% NaCl solution at room temperature (18-19°C) is 21 mg% (<math>9.9 \times 10^{-4} \text{ mol dm}^{-3}</math>, compiler).</p>	
<b>AUXILIARY INFORMATION</b>	
<b>METHOD/APPARATUS/PROCEDURE:</b> <p>After standing for more than two days the soln of sulfamethylisothiurea was filtered and the sulfonamide was assayed in the filtrate colorimetrically by the method of Druey and Oesterheld (1).</p>	<b>SOURCE AND PURITY OF MATERIALS:</b> <p>Nothing specified.</p> <hr/> <b>ESTIMATED ERROR:</b> <p>Nothing specified.</p> <hr/> <b>REFERENCES:</b> <p>1. Druey, J.; Oesterheld, G. <i>Helv. Chim. Acta</i> <u>1942</u>, <i>25</i>, 753.</p>