

COMPONENTS:		ORIGINAL MEASUREMENTS:	
(1) Manganese, bis(4-amino-N-2-thiazolyl-benzenesulfonamidato-N ^N ,0)-hydrate; C ₁₈ H ₁₆ MnN ₆ O ₄ S ₄ ·nH ₂ O; [84812-77-1]		Tskitishvili, M.G.; Shvelashvili, A. E.;	
(2) Hydrochloric acid; HCl; [7647-01-0]		Mikadze, I. I.; Zhorzholiani, N. B.;	
(3) Water; H ₂ O; [7732-18-5]		Chrelashvili, M. V. <i>Izv. Akad. Nauk Gruz. SSR, Ser. Khim.</i> <u>1981</u> , 7(4), 300-4.	
VARIABLES:		PREPARED BY:	
pH		R. Piekos	
EXPERIMENTAL VALUES:			
	Concentration of HCl (mol/l)	pH	10 ⁹ K _{SO} at 25°C
	5.0 × 10 ⁻³	5.54	1.50
	2.5 × 10 ⁻³	5.57	1.48
	1.0 × 10 ⁻³	5.65	1.52
	5.0 × 10 ⁻⁴	5.79	1.46
	2.5 × 10 ⁻⁴	6.08	1.49
	1.0 × 10 ⁻⁴	6.29	1.51
	5.0 × 10 ⁻⁵	6.45	1.49
	1.5 × 10 ⁻⁵	6.72	1.47
		Mean	1.49
AUXILIARY INFORMATION			
METHOD/APPARATUS/PROCEDURE:		SOURCE AND PURITY OF MATERIALS:	
The earlier described apparatus and method was used (1): in a glass vessel a mixt of 100 ml of HCl of appropriate concn and the solute were placed and shaken for 6 h in a water thermostat at 25°C. After attaining equilibrium, the pH of the soln was measured and the Mn ²⁺ and S content was detd to calculate K _{SO} . The pH was measured on a pH-673 pH meter.		0.1M solns of chemically pure Mn(OAc) ₂ , monosodium salt of sulfathiazole and HCl as well as doubly distd water were used. The source of the materials was not specified.	
		ESTIMATED ERROR:	
		K _{SO} : std deviation 2 × 10 ⁻¹¹ (compiler) Temp and pH: not specified.	
		REFERENCES:	
		1. Tskitishvili, M. G.; Mikadze, I. I.; <i>Soobshch. Akad. Nauk Gruz. SSR</i> <u>1978</u> , 89(3), 589.	