OMPONENTS :		ORIGINAL MEASUREMENTS:	
(1) 3,3-Dimethylpentane; C <sub>7</sub> H <sub>16</sub> ;		Price, L.C.	
[562-49-2] / 10		Am. Assoc. Petrol. Geol. Bull.	
(2) Water; H <sub>2</sub> O; [7732-18-5]		<u>1976</u> , 60, 213-44.	
ARIABLES:		PREPARED BY:	
Temperature: 25-150.4°C		F. Kapuku	
XPERIMENTAL VALUES	:	I	
Solubility	of 3,3-dimethylpentane	in water at system p	oressure
t/°C	mg(l)/kg(2)	g(l)/l00 g sln (compiler)	$10^{6}x_{1}$ (compiler)
25.0	5.92 ± 0.06	0.000592	1.06
40.1	6.78 ± 0.20	0.000678	1.22
55.7	8.17 ± 0.46	0.000817	1.47
69.7	10.3 ± 0.7	0.00103	1.85
99.1	15.8 ± 0.7	0.00158	2.84
118.0	27.3 ± 0.4	0.00273	4.91
140.4	67.3 ± 1.7	0.00673	12.10
150.4	86.1 ± 1.8	0.00861	15.48
	AUXILIARY		
ETHOD/APPARATUS/PR		INFORMATION	
Room-temperature solubilities were determined by use of screw-cap test tubes. The (1) phase floated on top of (2) and insured saturation (in 2 to 4 days) of the aqueous phase. High-temperature solubility work was		SOURCE AND PURITY OF MATH	ERIALS :
Room-temperatur determined by u tubes. The (1) of (2) and insu to 4 days) of t High-temperatur carried out in	CCEDURE: ce solubilities were use of screw-cap test phase floated on top ured saturation (in 2 the aqueous phase. ce solubility work was the ovens of the gas	SOURCE AND FURITY OF MATH (1) Phillips Petrole 99+%. (2) distilled.	ERIALS; eum Company;
Room-temperatur determined by u tubes. The (1) of (2) and insu to 4 days) of t High-temperatur carried out in chromatograph. contained in 75 stainless steel	CCEDURE: ce solubilities were use of screw-cap test phase floated on top ured saturation (in 2 the aqueous phase. ce solubility work was the ovens of the gas The solutions were mL double ended sample cylinders	Source AND PURITY OF MATH (1) Phillips Petrole 99+%. (2) distilled.	ERIALS: eum Company;
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(1) 3,3-Dimethyipentane; C7 <sup>H</sup> 16; [562-49-2]	Krzyzanowska, T.; Szeliga, J.				
(2) Water; H <sub>2</sub> O; [7732-18-5]	Nafta (Katowice) <u>1978</u> , 12, 413-7.				
	-				
VARIABLES:	PREPARED BY:				
One temperature: 25°C	M.C. Haulait-Pirson				
-					
EXPERIMENTAL VALUES:					
The solubility of 3,3-dimethylpentane in water at 25°C was reported					
to be 5.94 mg(l)/kg(2).					
The corresponding mass percent and mole fraction, $x_1$ , calculated					
by compiler are 5.94 x $10^{-4}$ g(l)/100 g sln and 1.07 x $10^{-6}$ .					
Editor's Note: Based on the results	for this and other hydrocarbon-water				
systems, uncertainity exists about w	nether the datum compiled here is				
independent of that of Price for the	same system (see previous page).				
Consequently, this system has not be	Consequently, this system has not been evaluated.				
AUXILIARY	INFORMATION				
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