

<b>COMPONENTS:</b>  (1) 1,4-Dicyclopentylbutane; $C_{14}H_{26}$ ; [2980-70-3]  (2) Water; $H_2O$ ; [7732-18-5]	<b>ORIGINAL MEASUREMENTS:</b>  Englin, B.A.; Plate, A.F.; Tugolukov, V.M.; Pryanishnikova, M.A.  <i>Khim. Technol. Topl. Masel</i> <u>1965</u> , 10, 42-6.												
<b>VARIABLES:</b>  Temperature: 10-30°C	<b>PREPARED BY:</b>  A. Maczynski and Z. Maczynska												
<b>EXPERIMENTAL VALUES:</b>  Solubility of Water in 1,4-Dicyclopentylbutane  <table border="1" data-bbox="220 574 963 725"> <thead> <tr> <th><math>t/^\circ C</math></th> <th><math>g(2)/100\ g\ sln</math></th> <th><math>10^3\ x_2</math> (compiler)</th> </tr> </thead> <tbody> <tr> <td>10</td> <td>0.0088</td> <td>0.94</td> </tr> <tr> <td>20</td> <td>0.0139</td> <td>1.50</td> </tr> <tr> <td>30</td> <td>0.0241</td> <td>2.59</td> </tr> </tbody> </table>		$t/^\circ C$	$g(2)/100\ g\ sln$	$10^3\ x_2$ (compiler)	10	0.0088	0.94	20	0.0139	1.50	30	0.0241	2.59
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<b>AUXILIARY INFORMATION</b>													
<b>METHOD/APPARATUS/PROCEDURE:</b>  Component (1) was introduced into a thermostatted flask and saturated for 5 hr. with (2). Next, calcium hydride was added and the evolving hydrogen volume measured and hence the concentration of (2) in (1) was evaluated.	<b>SOURCE AND PURITY OF MATERIALS:</b>  (1) Not specified. (2) Not specified.  <b>ESTIMATED ERROR:</b>  Not specified.  <b>REFERENCES:</b>												